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(21) International Application Number: PCT/IB99/01287 (22) International Filing Date: 14 July 1999 (14.07.99) (30) Priority Data: 9815269.7 14 July 1998 (14.07.98) GB (71) Applicant (for all designated States except US): ROLIC AG [CH/CH]; Innere Güterstrasse, CH-6301 Zug (CH). (72) Inventors; and (75) Inventors/Applicants (for US only): CHERKAOUI, Zoubair [MA/CH]; Dürrenmattweg 9, CH-4123 Allschwil (CH). BENECKE, Carsten [DE/DE]; Vierthausen 9, D-79576 Weil am Rhein (DE). (74) Agent: ENGLAND, Christopher, David; BTG International Ltd, 10 Fleet Place, Limeburner Lane, London EC4M 7SB (GB).		(81) Designated States: AE, AI., AM, AT, AU, AZ, BA, BB, BG, BR, BY, CA, CH, CN, CU, CZ, DE, DK, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MD, MG, MK, MN, MW, MX, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, UA, UG, US, UZ, VN, YU, ZA, ZW, ARIPO patent (GH, GM, KE, LS, MW, SD, SL, SZ, UG, ZW), Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG). Published <i>With international search report.</i>
(54) Title: COMPOSITIONS (57) Abstract Compounds are provided for use in the preparation of polymerisable liquid crystalline mixtures, liquid crystalline polymers and optical devices prepared from such mixtures and polymers. The component includes novel compounds comprising an organosiloxane portion and a portion having a structure that is compatible with a mesogenic molecular architecture.		

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Compositions

The present invention relates to the use of a component of a polymerisable liquid crystalline mixture in the production of orientated liquid crystalline polymers; compounds used as components in polymerisable liquid crystalline mixtures; liquid crystalline mixtures comprising these components, liquid crystalline polymers prepared from such components and liquid crystalline devices comprising those compounds.

Liquid crystal polymers (LCPs) are used in the manufacture of optical components such as waveguides, optical gratings, filters, retarders, piezoelectric cells and non-linear optical cells and films. The choice of LCP for use in any one of the aforementioned optical components depends upon its associated optical properties such as the optical anisotropy, refractive index, transparency and dispersion. Optical filters, for example, contain LCPs having a large anisotropy (Δn) and a low dispersion ($n=f(\lambda)$).

In some applications there is a requirement to produce LCPs in which the component molecules adopt a specific tilt angle with respect to the plane of the substrate. These LCP materials can be used as optical components such as compensation layers and retarders. Such optical components may be used in the production of liquid crystal devices (LCDs) with improved viewing angles, for example.

LCPs are manufactured by orientating a layer of a polymerisable liquid crystal single compound or mixture on an orientated substrate and cross-linking the mesogenic layer to form a liquid crystal polymer (LCP) network. Polymerisable LC compounds used in the manufacture of the LCPs need to be chemically and thermally stable, stable to electromagnetic radiation, soluble in standard solvents, miscible with other LC components and exhibit liquid crystalline properties over the range 25 to 150°C, preferably 25 to 80°C. The configuration imposed by the orientation layer on the polymerisable LC single compound or mixture becomes fixed or frozen into the LCP network formed upon cross-linking. The resulting LCP films have a high viscosity and are stable to mechanical stresses, temperature and light exposure.

The tilt angle adopted by the polymerisable LC single compound or mixture prior to cross-linking depends in part upon the nature of the LC components comprising that single compound or mixture. Previous strategies employed in generating a tilt angle in a given LC single compound or mixture prior to cross-linking have, for example, relied

upon the introduction of one or more lateral substituents into the mesogen, such as a short aliphatic alkyl chain. However, such substitution is accompanied by a decrease in the mechanical properties of the LC material or, depending upon the molecular architecture of the mesogen, is associated with a small or vanishing tilt angle. In each case the reproducibility of the initial tilt is readily lost before the mixture is cross-linked to give the LCP network due to thermal and mechanical effects.

There is therefore a need for a liquid crystalline single compound or mixture that can be orientated on a substrate prior to cross-linking in such a way that the orientation of the LC single compound or mixture on the substrate remains stable over the period required for manufacturing the LCP network. There is also a need for a LC component with the ability to facilitate the generation of and maintain a wide range of predetermined tilt angles prior to and during the manufacture of the LCP film. The present invention addresses those needs.

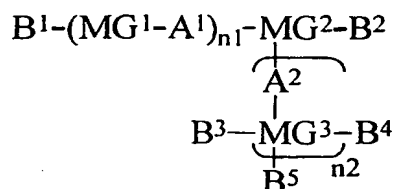
A first aspect of the present invention provides the use of a component of a LC mixture in the manufacture of a LCP network comprising the mixture, characterised in that the component induces a stable tilt angle in the LCP network. By the word stable it is to be understood to mean that there is no substantial change in the value of the tilt angle over the period necessary for manufacturing the LCP network. The LCP network obtained by cross-linking the polymerisable LC mixture can be a nematic, smectic A or discotic film, but is preferably a nematic film. These films are characterised by good orientation of the film components relative to the substrate.

The component used to induce a stable tilt angle is preferably mesogenic in nature or is compatible with a mesogenic molecular architecture and may contain polymerisable or non-polymerisable substituents. It is preferably miscible with the other components of the liquid crystal mixture. Without wishing to limit the scope of the invention, it is believed that the component must contain a sterically bulky group that has the ability to form a strong association with the substrate or with the orientation layer. The component should also contain a sterically less bulky group that is compatible with a mesogenic molecular architecture and which extends into the bulk of the LC single compound or mixture as a whole. It is believed that this mesogenically compatible group is able to assume an angle in which it is tilted relative to the plane of the substrate. Unlike the LCP

mixtures of the prior art, LCPs including these components are less susceptible to thermal effects and the polymerisable LC single compound or mixture comprising or including a component of the present invention is able to retain its initial orientation over the period necessary for manufacturing the LCP network. Judicious choice of the substrate associating and mesogenic extending groups allows this angle to be varied. It will be appreciated that the components of the invention can therefore be used in the manufacture of polymerisable LC single compounds or mixtures having a stable orientation or tilt angle relative to the plane of the substrate.

By sterically bulky it is to be understood that the lateral dimensions of the substrate associating group are at least half as large again as the lateral dimensions of the less bulky group, preferably two or more times the dimensions of the less bulky group.

The substrate associating or more bulky groups preferably comprise organosiloxane groups and the mesogenic extending or less bulky groups preferably contain aromatic or non-aromatic carbocyclic or heterocyclic ring systems. It is believed that the organosiloxane containing components used in the first aspect of the invention include compounds that are new *per se* and a second aspect of the invention provides an organosiloxane compound of formula I



(I)

wherein

B¹ to B⁴ each individually represent a spacer group selected from the group comprising optionally substituted saturated or unsaturated, straight chain or branched chain C₂₋₈₀ alkyl group, one or more of the non-adjacent alkyl CH₂ groups being optionally replaced by one or more heteroatoms;

B⁵ represents a hydrogen atom or spacer group as defined above,

A¹ and A² each individually represent a single bond or a spacer group as defined above,

MG¹ to MG³ each individually represent a mesogenic group comprising at least one optionally substituted aromatic or non aromatic carbocyclic or heterocyclic ring systems; and

n₁ and n₂ are each independently 0 or a positive integer

with the proviso that, firstly, when n₂ is 0 both B¹ and B² include a polymerisable group and at least one of A¹, B¹ and B² includes a group selected from an organosiloxane, an organogermanium, an organotin and an organo-perfluoro residue and secondly, when n₂ > 0 at least one of B¹, B², B³ and B⁴ includes a polymerisable group and at least one of A¹, A², B¹, B², B³ and B⁴ includes an organosiloxane group.

By the term 'heteroatom' it should be understood to include nitrogen, oxygen and sulphur. Replacement nitrogen atoms may be further substituted with groups such as alkyl, aryl and cycloalkyl. These compounds have been found to be particularly useful in the preparation of liquid crystalline polymer networks (LCPs) in which the molecules are tilted with respect to the orientation layer. These LCPs can be used in a range of different optical and electrical applications. LCP networks having nematic, smectic A and surprisingly discotic properties may be prepared. Nematic mesophases are preferred.

Most of the discotic liquid crystalline compounds described in prior art are based on disc-shaped molecules (such as triphenylene) which are characterised by a rigid and highly conjugated aromatic central core. These discotic compounds are of limited use because of properties such as a high viscosity, a high melting point, near visible light absorption and problems in their preparation, which also hinders the chemical diversity of the compounds that may be prepared.

There is, therefore, a need for new discotic liquid crystalline materials having reduced melting points and viscosities compared to those obtained with calamitic liquid crystals.

The present inventors have surprisingly found that discotic mesophases may be obtained from rod-shaped molecules, which are more easily prepared and offer the possibility of preparing material having diverse molecular structures. These discotic mesophases may be obtained from molecules having calamitic architecture, by

introducing into each one of their peripheral alkyl chains, at least one bulky group which is not compatible (in the liquid crystalline state), with the other molecular constituents. This non compatibility means that these bulky groups tend to associate in the mesophase leading to separate micro layers.

5 The presence of the bulky groups in each of the peripheral alkyl chains causes the central rigid rod-shaped cores to adopt an intermolecular crossed configuration leading to disc-shaped elements which are able to form discotic mesophases.

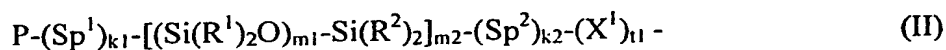
 The compound of formula (I) may be monomeric, oligomeric or polymeric in nature. It will be appreciated that the value of the integers n_1 and n_2 will be determined
10 by the nature of the compound of formula (I). When the compound of formula (I) is monomeric or oligomeric in nature, n_1 and n_2 may assume values of from 0 to 10, preferably from 1 to 5. However when the compound of formula (I) is polymeric in nature the values of n_1 and n_2 can range from 6 to 3,000 preferably from 11 to 2,000.

 The number of aromatic or non-aromatic ring systems present in the groups MG^1
15 to MG^3 will depend upon the application in which the compound of formula (I) is to be used. MG^1 to MG^3 preferably contain between 1 and 4 aromatic or non-aromatic ring systems.

 The hydrocarbon spacer groups B^1 to B^4 may include C_1 - C_{20} -alkyl, C_1 - C_{20} -alkoxy, C_1 - C_{20} -alkoxycarbonyl, C_1 - C_{20} -alkylcarbonyl and C_1 - C_{20} -alkylcarbonyloxy groups, for
20 example, methyl, ethyl, propyl, butyl, pentyl, hexyl, heptyl, octyl, nonyl, decyl, undecyl, dodecyl, methoxy, ethoxy, n-propoxy, i-propoxy, butoxy, pentyloxy, hexyloxy, heptyloxy, octyloxy, nonyloxy, decyloxy, undecyloxy, dodecyloxy, methoxycarbonyl, ethoxycarbonyl, propoxycarbonyl, butoxycarbonyl, pentyloxycarbonyl, hexyloxycarbonyl, octyloxycarbonyl, nonyloxycarbonyl, decyloxycarbonyl, undecyloxycarbonyl, dodecyloxycarbonyl,
25 methylcarbonyl, ethylcarbonyl, propylcarbonyl, butylcarbonyl, pentylcarbonyl, hexylcarbonyl, octylcarbonyl, nonylcarbonyl, decylcarbonyl, undecylcarbonyl, dodecylcarbonyl, methylcarbonyloxy, ethylcarbonyloxy, propylcarbonyloxy, butylcarbonyloxy, pentylcarbonyloxy, hexylcarbonyloxy, octylcarbonyloxy, nonylcarbonyloxy, decylcarbonyloxy, undecylcarbonyloxy, dodecylcarbonyloxy, and the
30 like.

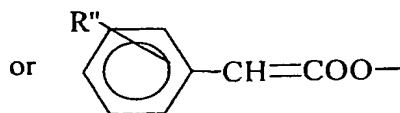
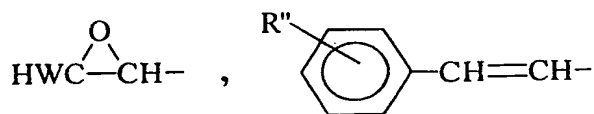
The optionally substituted C_{2-80} alkyl groups may be substituted by alkyl, aryl and cycloalkyl, as well as amino, cyano, epoxy, halogen, hydroxy, nitro, oxo etc. Possible heteroatoms, which may replace carbon atoms, include nitrogen, oxygen and sulfur. In the case of nitrogen further substitution is possible with groups such as alkyl, aryl and cycloalkyl. The optionally substituted aromatic or non-aromatic carbocyclic or heterocyclic ring systems may be similarly substituted.

Any one of the groups B^1 , B^2 , B^3 and B^4 may contain an organosiloxane group and a polymerisable group. Alternatively one or more of the groups B^1 , B^2 , B^3 , B^4 , A^1 and A^2 may contain an organosiloxane group with one or more of the other groups of B^1 , B^2 , B^3 and B^4 containing a polymerisable group. Preferably at least one of the groups B^1 to B^4 includes both a polymerisable group and an organosiloxane group and in a first preferred embodiment of the second aspect of the invention each or any of the groups B^1 to B^4 may be a hydrocarbon residue of formula (II)



wherein

P is a polymerisable group selected from groups comprising $CH_2=CW-$, $CH_2=W-O-$, $CH_2=CW-COO-$, $CH_2=C(Ph)-COO-$, $CH_2=CH-COO-Ph$, $CH_2=CW-CO-NH-$, $CH_2=C(Ph)-CONH-$, $CH_2=C(COOR')-CH_2-COO-$, $CH_2=CH-O-$, $CH_2=CH-OOC-$, $Ph-CH=CH-$, $CH_3-C=N-(CH_2)_{m3}-$, $HO-$, $HS-$, $HO-(CH_2)_{m3}-$, $HS-(CH_2)_{m3}-$, $HO(CH_2)_{m3}COO-$, $HS(CH_2)_{m3}COO-$, $HWN-$, $HOC(O)-$, $CH_2=CH-Ph-(O)_{m4}$



wherein

W represents H, F, Cl, Br or I or a C₁₋₅ alkyl group;

m₃ is an integer having a value of from 1 to 9;

m₄ is an integer having a value of 0 or 1,

5 R' represents a C₁₋₅ alkyl group;

R'' represents a C₁₋₅ alkyl group, methoxy, cyano, F, Cl, Br or I;

Sp¹ and Sp² each independently represent a C₁₋₂₀ alkyl group;

k₁ and k₂ are integers, each independently having a value of from 0 to 4;

R¹ and R² each independently represent straight chain or branched C₁₋₂₀ alkyl groups;

10 m₁ is an integer having a value of from 0 to 10;

m₂ is an integer having a value of from 1 to 10;

X¹ represents -O-, -S-, -NH-, N(CH₃)-, -CH(OH)-, -CO-, -CH₂(CO)-, -SO-,
-CH₂(SO)-, -SO₂-, -CH₂(SO₂)-, -COO-, -OCO-, -OCO-O-, -S-CO-, -CO-
S-, -SOO-, -OSO-, -SOS-, -CH₂-CH₂-, -OCH₂-, -CH₂O-, -CH=CH-, -C≡C-

15 or a single bond; and

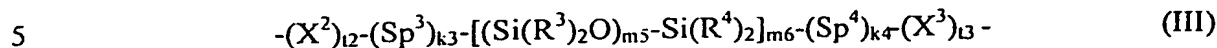
t₁ is an integer having a value of 0 or 1.

In relation to the hydrocarbon residue of formula (II), the term -Ph- is to be understood as representing 1,4-phenylene, 1,2-phenylene as well as 1,3-phenylene. 1,4-phenylene groups are preferred. The term Ph- or (Ph) should be understood to represent a
20 phenyl group.

The C₁₋₂₀ alkyl groups Sp¹ and Sp² may comprise branched or straight chain alkyl groups and may be unsubstituted, mono- or polysubstituted by F, Cl, Br, I or CN. Alternatively or in addition one or more of the CH₂ groups present in the hydrocarbon chain may be replaced, independently, by one or more groups selected from -O-, -S-,
25 -NH-, N(CH₃)-, -CH(OH)-, -CO-, -CH₂(CO)-, -SO-, -CH₂(SO)-, -SO₂-, -CH₂(SO₂)-, -COO-, -OCO-, -OCO-O-, -S-CO-, -CO-S-, -SOO-, -OSO-, -SOS-, -C≡C-, -(CF₂)_r, -CD₂)_s- or C(W¹)=C(W²)-, with the proviso that no two oxygen atoms are directly linked to each other. W¹ and W² each represent, independently, H, H-(CH₂)_{q1}-, F, Cl, Br or I. The integers r, s and q₁ each independently represent a number of between 1 and 15.

30 One or more of the CH₂ groups of the C₁₋₂₀ alkyl groups R¹ and R² may each be replaced, independently, by -Si(CH₃)₂- or / and by -Si(CH₃)₂O Si(CH₃)₂-.

Compounds of formula (I) in which A¹ and/or A² comprise an organosiloxane group are preferred and in a second preferred embodiment of the second aspect of the invention the groups A¹ and/or A² comprise a group of formula (III)

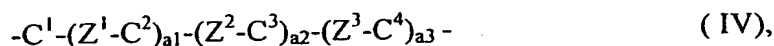


wherein

- Sp³ and Sp⁴ each independently represent the designations assigned to Sp¹ and Sp² above,
- 10 X² and X³ each independently represent the designations assigned to X¹ above,
R³ and R⁴ each independently represent the designations assigned to R¹ and R² above,
- k₃ and k₄ are integers, each integer independently having a value of from 0 to 4,
l₂ and l₃ are integers, each independently having a value of 0 or 1; and
- 15 m₅ is an integer having a value of from 0 to 10; and
m₆ is an integer having a value of from 1 to 10.

The groups MG¹ to MG³ may be the same or different and may comprise up to three bridging groups. A third embodiment of the second aspect of the invention provides

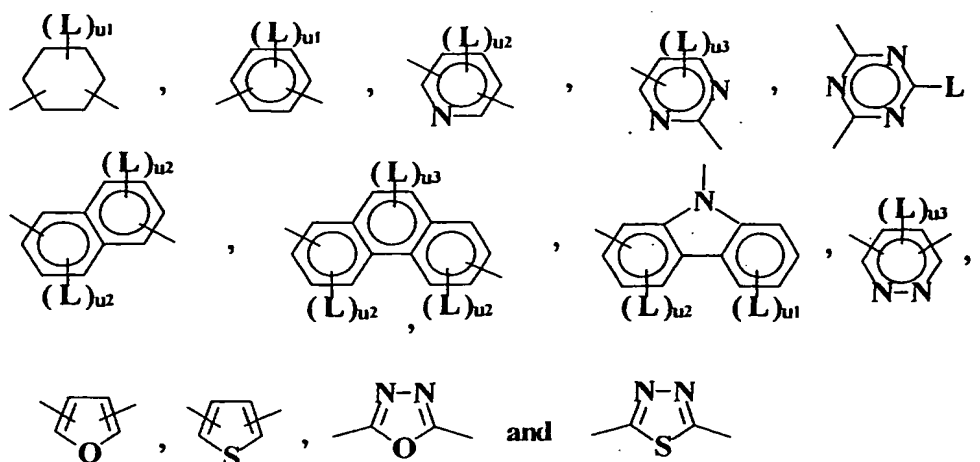
20 compounds of formula (I) in which MG¹ comprises a group of formula (IV)



wherein

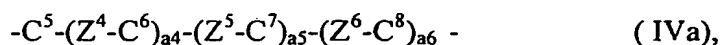
- C¹ to C⁴ each independently represent a non aromatic, aromatic, carbocyclic or
- 25 heterocyclic group;
- Z¹ to Z³ each independently represent -COO-, -OCO-, -CH₂-CH₂-, -OCH₂-,
-CH₂O-, -CH=CH-, -C≡C-, -CH=CH-COO-, -OCO-CH=CH- or a single
bond; and
- a₁, a₂ and a₃ are integers, each integer independently having a value of from 0 to 3,
- 30 with the proviso that a₁ + a₂ + a₃ ≤ 3.

It is especially preferred that the groups C^1 to C^4 are selected from the following structural formulae:



5

In a fourth embodiment of the second aspect of the invention the groups MG^2 and MG^3 comprise groups of formula (IVa)

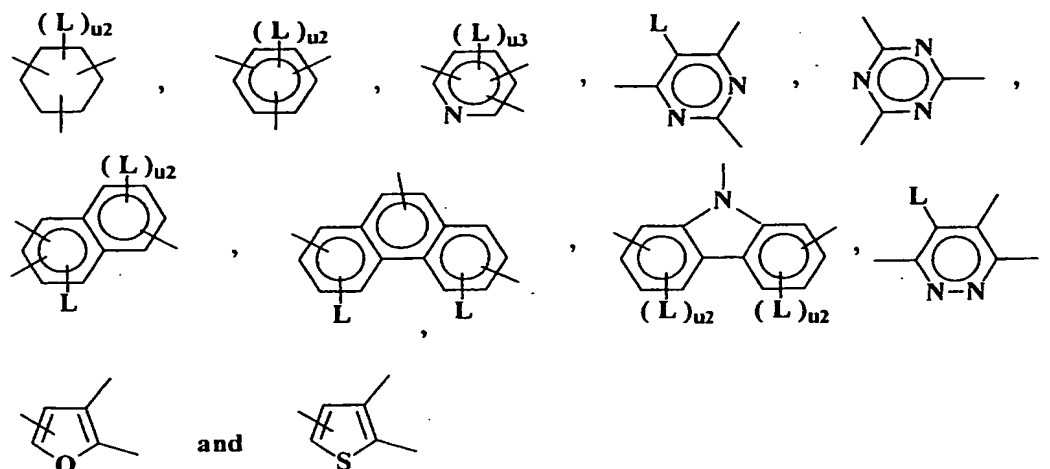


wherein:

- 10 C^5 , C^6 , C^7 and C^8 each independently represent a non-aromatic, aromatic, carbocyclic or heterocyclic group; and
- Z^4 to Z^6 each independently represent $-COO-$, $-OCO-$, $-CH_2-CH_2-$, $-OCH_2-$, $-CH_2O-$, $-CH=CH-$, $-C\equiv C-$, $-CH=CH-COO-$, $-OCO-CH=CH-$ or a single bond; and
- 15 $a4$, $a5$ and $a6$ are integers, each integer independently having a value of from 0 to 3,

with the proviso that firstly $a4 + a5 + a6 \leq 3$ and secondly that at least one of C^5 to C^8 independently comprises a trivalent group.

- 20 It is preferred that C^7 is the trivalent group. It is especially preferred that C^7 selected from one of the following structural formulae:



wherein for each of the groups MG^1 to MG^3 of formulae (IV) and (IVa) respectively

- 5 L represents $-CH_3$, $-COCH_3$, $-NO_2$, CN, or halogen;
 u1 represents an integer having a value of from 0 to 4;
 u2 represents an integer having a value of from 0 to 3; and
 u3 represents an integer having a value of from 0 to 2.

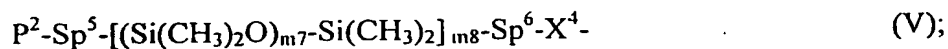
10 In an especially preferred fifth embodiment of the second aspect of the invention there is provided a compound of formula (I)

wherein

n_1 and n_2 are independently 0 or 1;

B^5 is hydrogen;

15 B^1 to B^4 each independently represent a group of formula (V)



A^1 and A^2 each independently represent a group of formula (VI)

20



wherein

X^4 to X^6 each independently represent -O-, -CO-, -COO-, -OCO-, -C≡C-, or a single bond, especially -O-, -COO-, -OCO- or single bond;

Sp^5 to Sp^8 each independently represent a C_{1-20} straight-chain alkyl group, especially ethylene, propylene, butylene, pentylene, hexylene, heptylene, octylene, nonylene, decylene, undecylene, or dodecylene;

p^2 represents $CH_2=CW^5$ - or $CH_2=CW^5-(CO)_{v2}O$ -

wherein

W^5 represents H, CH_3 , F, Cl, Br or I; and

$v2$ is 0 or 1;

$m7$ and $m9$ are integers, each independently having a value of 1 or 2;

$m8$ and $m10$ are integers, each independently having a value of 0 or 1;

with the proviso that $m8 + m10 > 0$;

15 MG^1 represents a group of formula IV

wherein

C^1 , C^2 , C^3 and C^4 are the same and are selected from phenylene or biphenylene;

and

Z^1 to Z^3 each independently represent a single bond, -COO- or -OCO-; and

20 MG^2 and MG^3 each independently represent a group of formula IVa

wherein

C^5 , C^6 , and C^8 are the same and are selected from phenylene or biphenylene;

C^7 represents phenylene or biphenylene; and

Z^4 to Z^6 each independently represent a single bond, -COO- or -OCO-.

The surprising ability of the compounds of the invention to form discotic mesophases has been discussed to above. The compounds of the invention that have been found to be particularly suitable for the preparation of discotic mesophases are those having a bulky group in each of their peripheral alkyl chains B¹ and B². As mentioned previously it is believed that the presence of the bulky groups causes the rod-shaped molecules of the invention to associate thereby forming disc-shaped elements, which are stabilising to discotic mesophases.

Suitable bulky groups include fluorinated alkyl residues as well as and organic residues having bulky atoms such as germanium, silicon, and tin.

Preferably the bulky groups comprise organic residues having tetravalent bulky atom such as germanium, silicon, and tin. It is especially preferred that the bulky groups comprise at least one siloxane group. It is more especially preferred that each of the groups B¹ and B², include at least one siloxane group.

It is preferred that the central core of molecules having the ability to form a discotic mesophase comprises at least one optionally substituted aromatic or non-aromatic carbocyclic ring system. More preferably the central cores are selected from those that are able to induce a calamitic mesophase.

The bulky groups are preferably each independently attached to the central core of the molecule via a C₁₋₂₀ alkyl spacer group as defined above.

A preferred sixth embodiment of the second aspect of the invention provides a compound of formula (I) in which

n₁ and n₂ are both 0,

B¹ and B² each independently represent a group of formula (VII)

$P^3-Sp^5-[Si(CH_3)_2O]_{m7}-Si(CH_3)_2]_{m8}-Sp^6-X^4-$ (VII);

wherein

X⁴ represents -O-, -CO-, -COO-, -OCO-, -C≡C- or a single bond, especially -O-, -COO-, -OCO- or single bond;

Sp⁵ to Sp⁶ each independently represent a C₁₋₂₀ straight-chain alkyl group, especially ethylene, propylene, butylene, pentylene, hexylene, heptylene, octylene, nonylene, decylene, undecylene, or dodecylene;

P³ represents a hydrogen or CH₂=CW⁵- or CH₂=CW⁵-(CO)_{v2}O-
wherein
W⁵ represents H, CH₃, F, Cl, Br or I; and
v2 is 0 or 1;

5 m7 is 1 or 2;

m8 is 1;

MG² represents a group of formula IVa
wherein

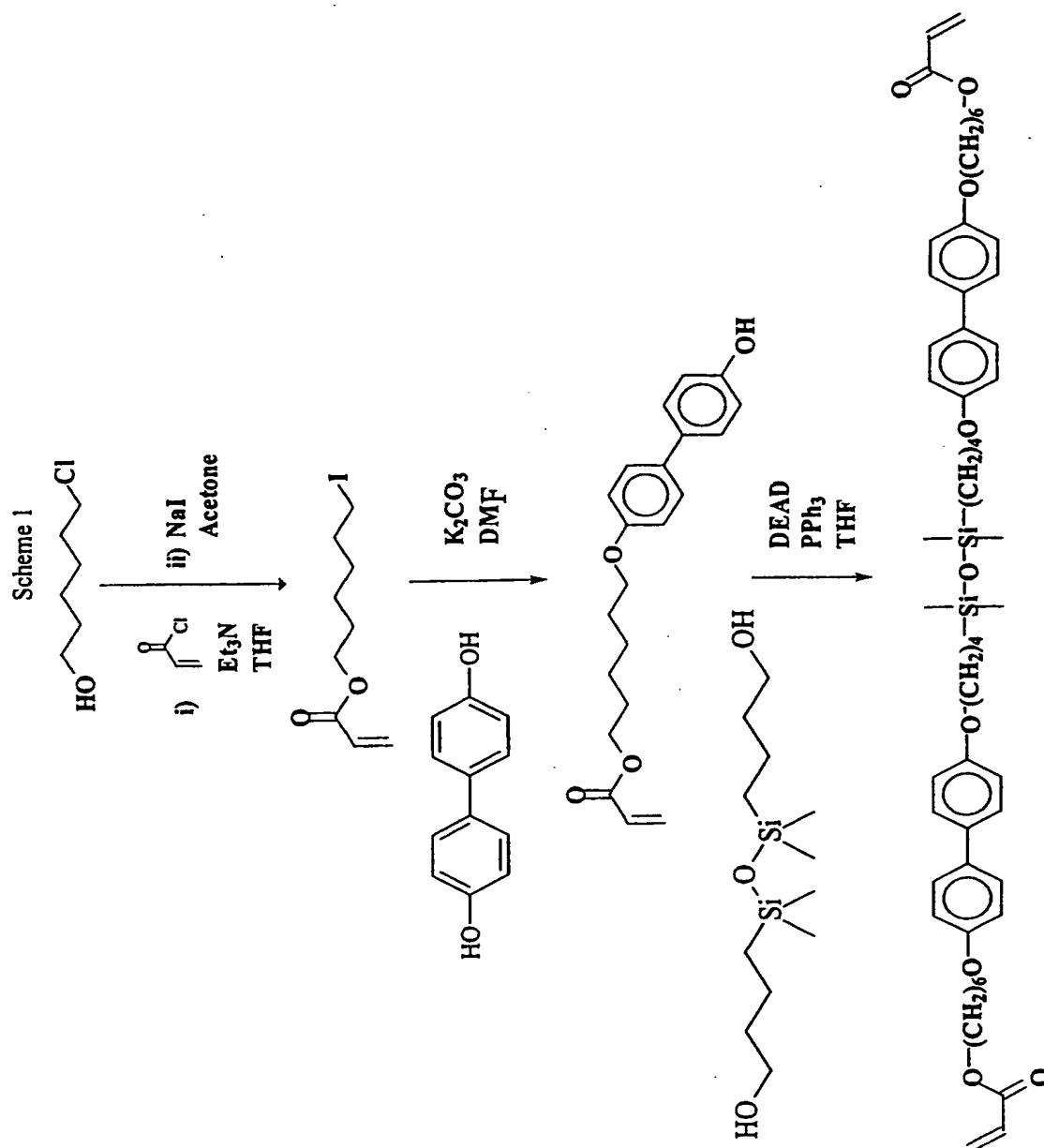
10 C⁵, C⁶, and C⁸ are the same and are selected from phenylene or
biphenylene;

C⁷ represents phenylene or biphenylene; and

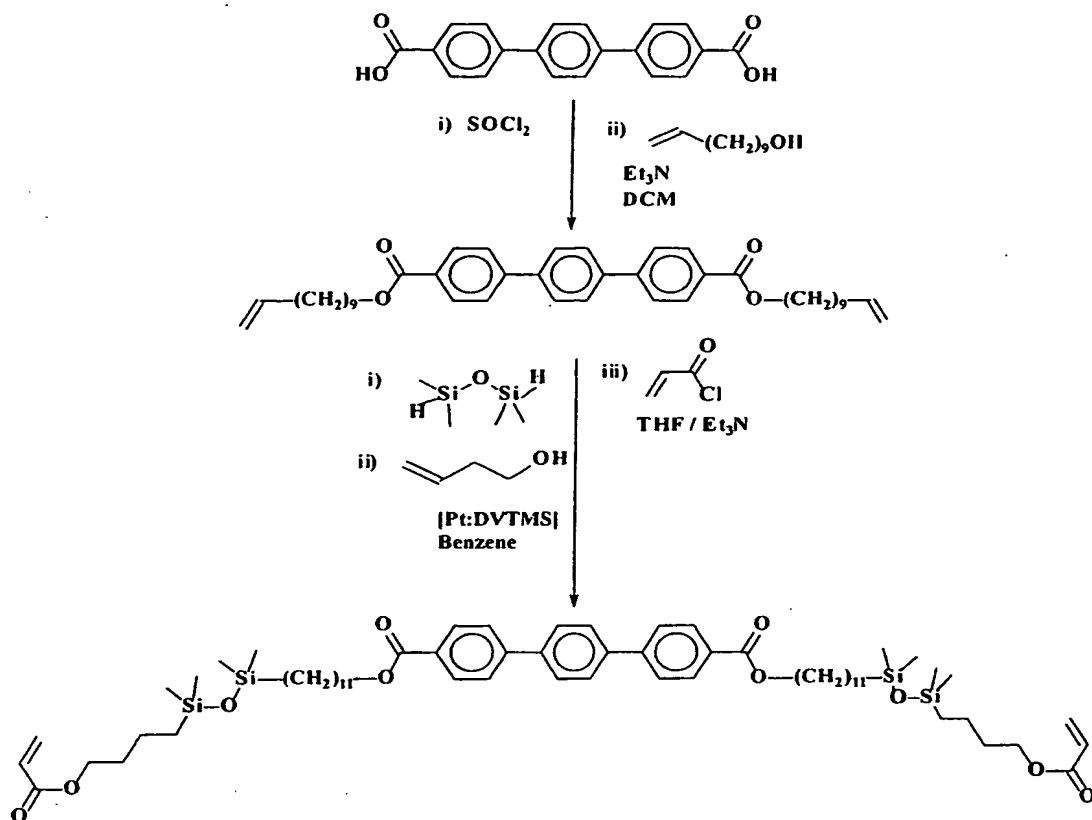
Z⁴ to Z⁶ each independently represent a single bond, -COO- or -OCO-.

15 These compounds are particularly suitable for the preparation of materials having
discotic mesophases.

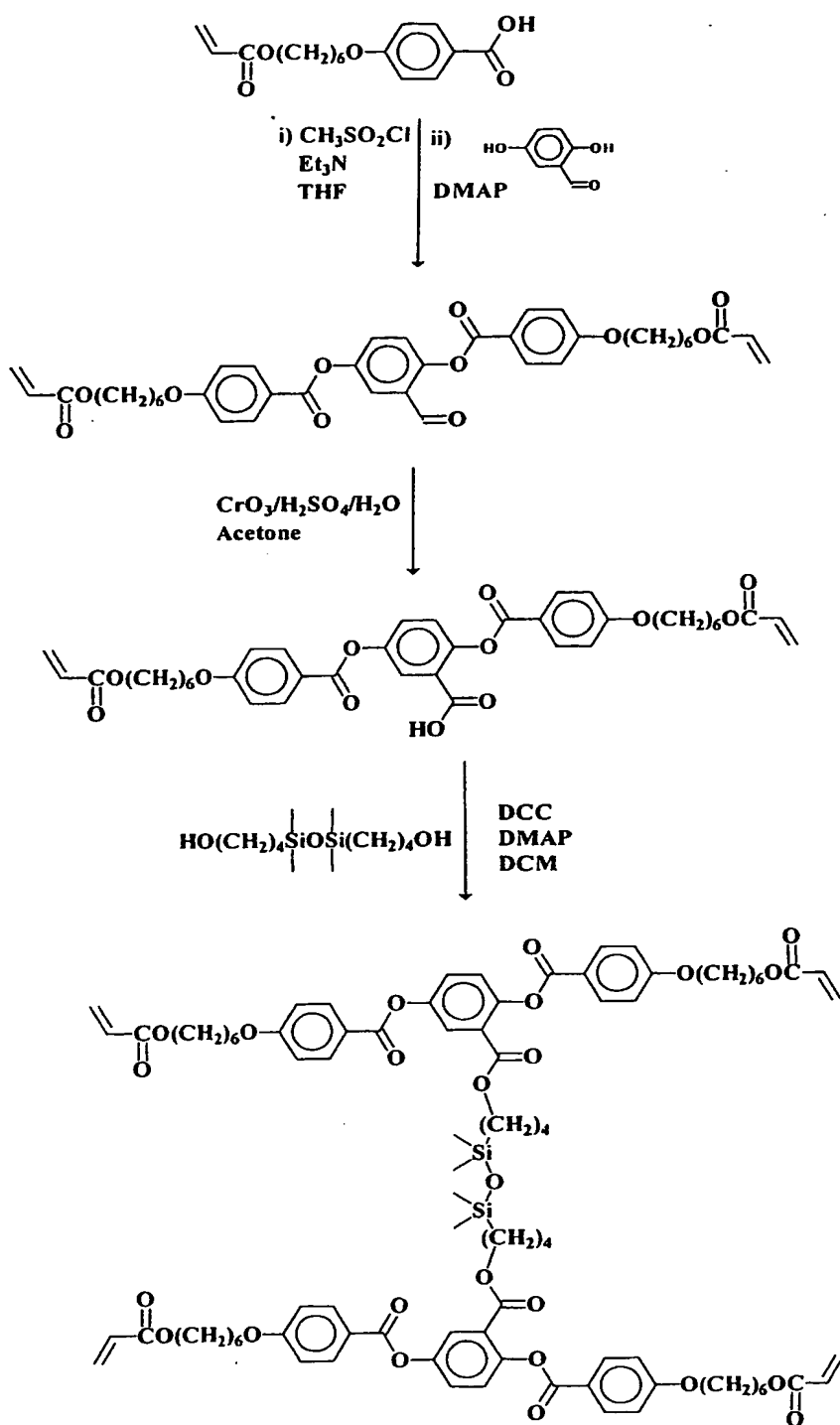
20 The compounds according to the second aspect of the invention may be readily
prepared using methods that are well known to the person skilled in the art, such as those
documented in Houben-Weyl, Methoden der Organischen Chemie, Thieme-Verlag,
Stuttgart. In particular, the compounds may be readily prepared from direactive
organosiloxane compounds according to any one of Schemes 1 to 5. A third aspect of the
invention provides a method for the preparation of compounds according to the second
aspect of the invention comprising reacting a direactive siloxane with a mesogenic
compound comprising at least one carbocyclic or heterocyclic ring system.



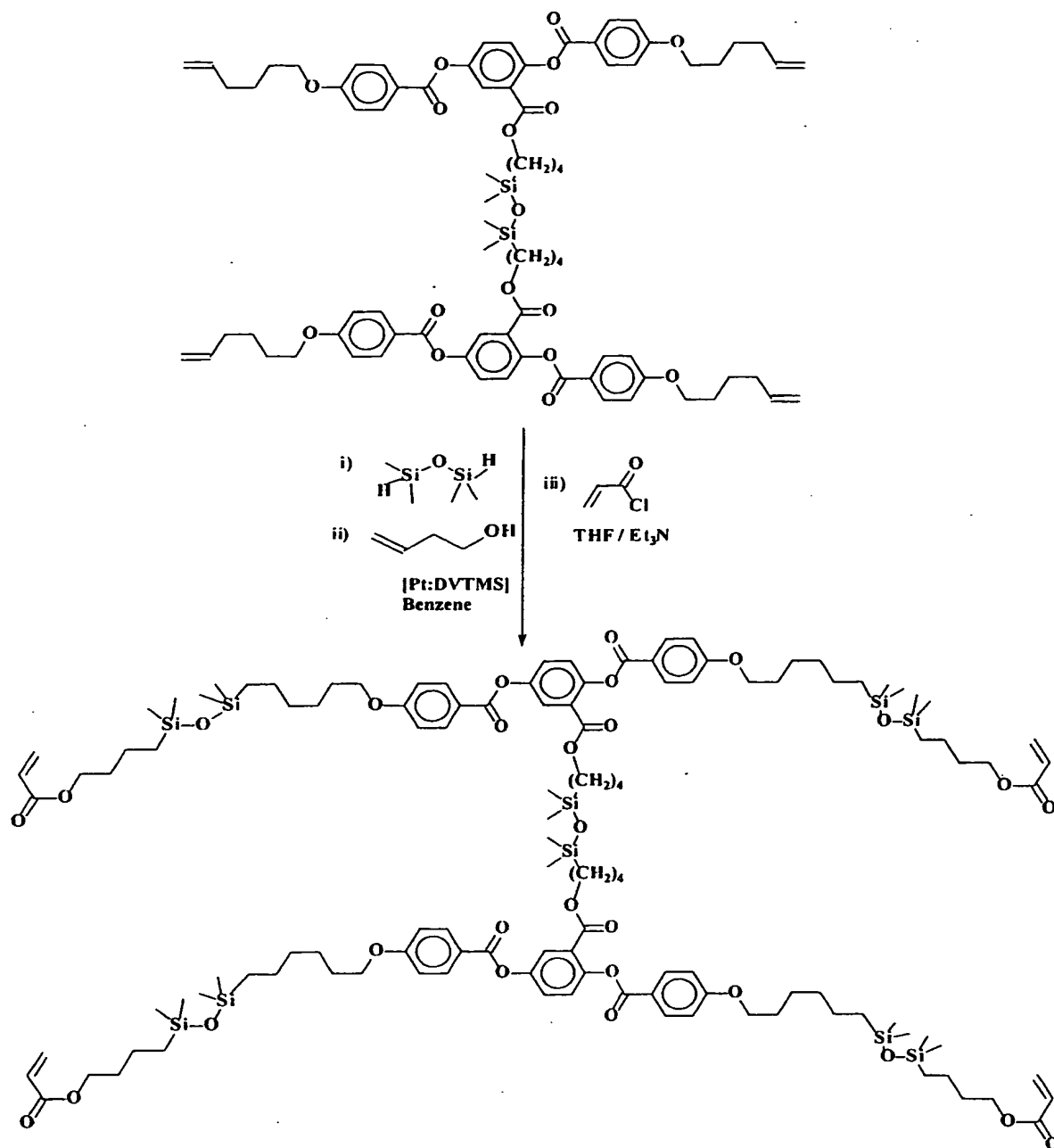
Scheme 2:



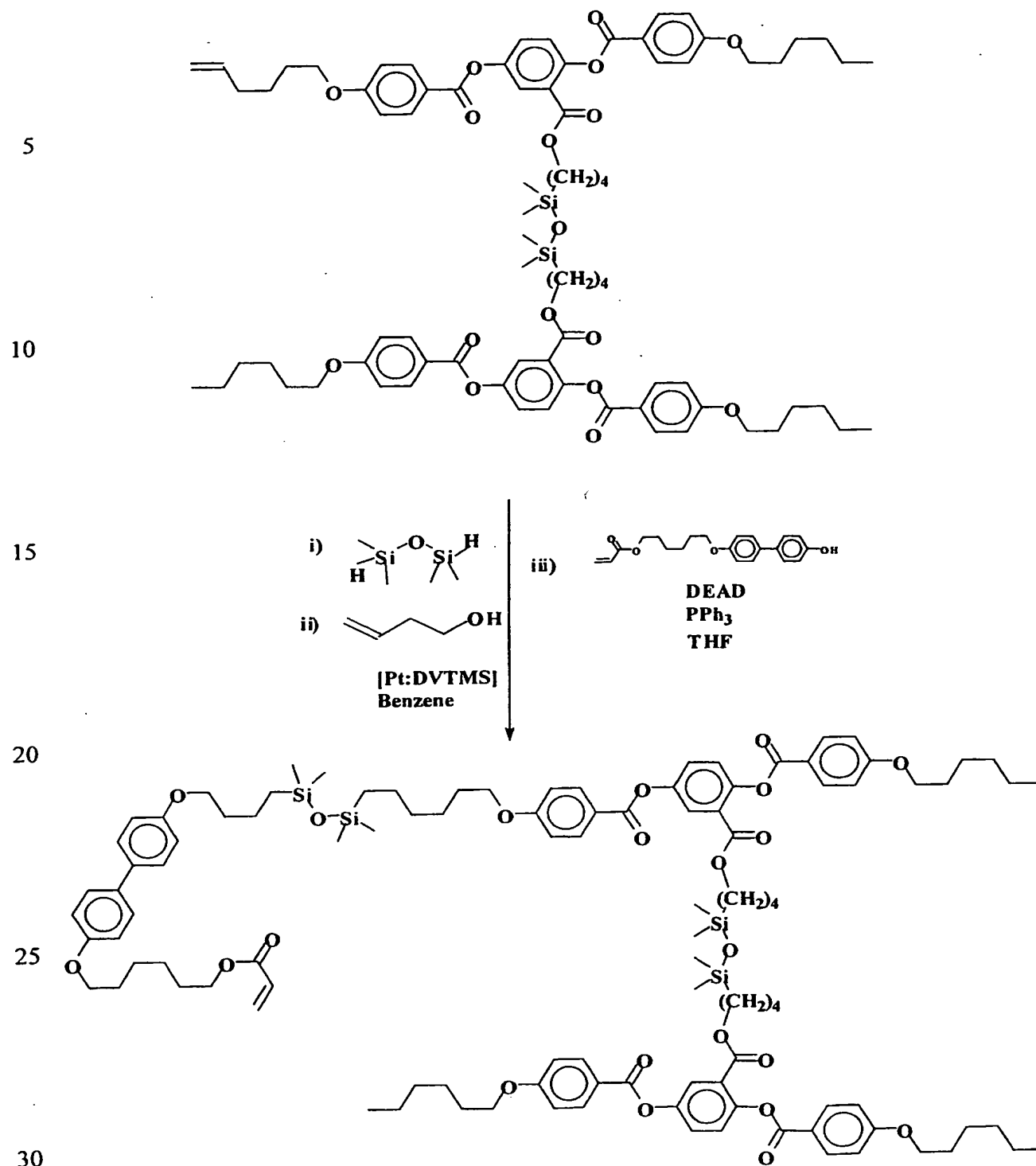
Scheme 3:



Scheme 4



Scheme 5



The organosiloxane compounds according to the second aspect of the invention may be used alone or as a component of a liquid crystal mixture. Liquid crystalline materials comprising a compound of formula (I) may be used in the manufacture of LCPs. A fourth aspect of the invention therefore comprises a liquid crystalline material comprising a compound of formula (I). Preferably the liquid crystalline materials comprises at least two components. The additional components must be miscible with the compound of formula (I) and may selected from known mesogenic materials such as those reported in Adv. Mater. 5, 107 (1993), Mol. Liq. Cryst. 307, 111 (1997), J. Mat. Chem. 5, 2047 (1995) or in patent applications US-A-5, 593, 617; US-A-5, 567, 349; GB-A-2 297 556; GB-A-2 299 333; DE-A-195 04 224; EP-A-0 606 940; EP-A-0 643 121 and EP-A-0 606 939, optionally selected from EP-A-0 606 940; EP-A-0 643 121 and EP-A-0 606 939.

The form of the liquid crystal material will depend upon the application in which it is to be used and may be present as a liquid crystalline mixture, (co)polymer, elastomer, polymer gel or polymer network. Polymer networks have been found to be of particular use and in a first preferred embodiment of the fourth aspect of the invention there is provided a polymer network comprising a compound of formula (I) in cross-linked form. Preferably the polymer network comprises at least two components, at least one of which is an organosiloxane compound of formula (I).

The polymer network may be prepared by the copolymerisation of a mesogenic mixture comprising:

- i) at least one chiral or/and achiral mesogenic polymerisable compound;
- ii) at least one organosiloxane compound of formula I; and
- iii) an initiator.

The chiral or achiral mesogenic polymerisable compound may be an organosiloxane compound of formula (I). Alternatively or in addition, the polymerisable compound may be selected from the known mesogenic materials referred to above. Preferably the chiral or achiral polymerisable compound includes the nematic phase in its thermotropic sequence.

The polymer network may optionally comprise further components. These include further polymerisable compounds, stabilisers and dyes. The additional polymerisable compounds preferably comprise a non-mesogenic compound having at least one polymerisable functional group, especially diacrylate compounds.

5 Any suitable stabiliser that prevents undesired spontaneous polymerisation, for example during storage of the mixture, may be used in the liquid crystalline mixture according to the invention. A broad range of these compounds is commercially available. Typical examples include 4-ethoxyphenol or 2,6-di-(t-butyl)-4-hydroxytoluene (BHT).

If colour filters are required, dyes may be added to the mixture. In a preferred
10 embodiment of the invention the LC mixture contains no dye.

The chiral or achiral polymerisable mesogenic compound may be present in an amount comprising 0.01 to 99% by weight of the liquid crystalline polymer network mixture, preferably 50 to 95% by weight.

The organosiloxane compound of formula (I) may be present in an amount from
15 0.1 to 100% by weight of the liquid crystalline network, preferably from 1 to 50% by weight.

The initiator is preferably a photoinitiator and may be a radical or cationic initiator that is present in an amount comprising 0.1 to 5% by weight of the polymer mixture, preferably from 0.2 to 2% by weight.

20 When the mixture further comprises a stabiliser, this is generally present in an amount comprising 0 to 5% by weight of the liquid crystalline mixture, preferably from 0.1 to 1% by weight.

These polymerisable liquid crystalline mixtures may be formed into liquid crystalline polymer (LCP) films and a fifth aspect of the invention provides a LCP film
25 comprising a compound of formula (I). LCP films may be readily prepared by UV polymerisation of a LC mixture according to the fourth aspect of the invention; a film comprising the LC mixture is formed on a substrate and polymerised using UV light to give a cross-linked liquid crystal polymer (LCP) film. The film is both light and temperature stable and can be used in the manufacture of devices such as waveguides,
30 optical gratings, filters, retarders, piezoelectric cells or thin films exhibiting non-linear optical properties.

Different methods can be used for the formation of the sought LCP network, starting from the polymerisable liquid crystalline mixture manufactured as described above. Transparent substrates such as coated ITO (indium tin oxide), glass or plastic substrates, may be used. Preferred substrates include glass or plastic, especially those including a layer of rubbed polyimide or polyamide or a layer of photooriented photopolymer (LPP). Said layers are used to facilitate uniform orientation of the liquid crystalline mixture.

In the preparation of LCP films, it is particularly important to prevent the formation of defects or inhomogenities. This can be achieved by forming the polymerisable liquid crystalline mixture into a thin film; placing the mixture between two of the aforementioned substrates which are then sheared over a small distance until a planar order was obtained; or capillary filling the polymerisable liquid crystalline mixture between two of the said substrates prior to curing with, for example, by UV light, preferably in the presence of a photoinitiator, such as Irgacure®.

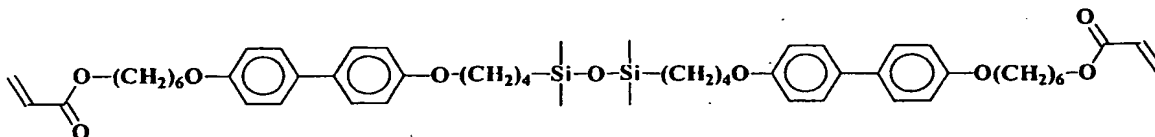
The invention will now be described with reference to the following examples. Variations on these falling within the scope of the invention will be apparent to a person skilled in the art.

In the following Examples the thermotropic phases are abbreviated as follows:

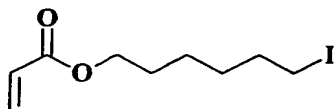
K	crystalline
D	discotic
S	smectic
N	nematic
N*	chiral nematic (cholesteric)
I	isotropic

Example 1:

1,3-Bis-{4-[4-oxo-4'-(6-acryloyloxyhexyloxy)biphenyl]but-1-yl}-1,1,3,3-tetramethyldisiloxane



(a) Acrylic acid 6-iodohexyl ester

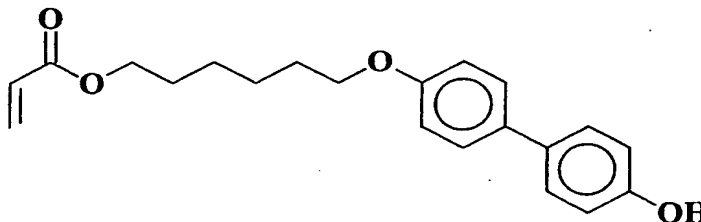


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To a stirred solution of 6-chloro-1-hexanol (27.3 g , 0.2 mol), triethylamine (56 ml , 400ml) in 100 ml of dichloromethane, cooled at 0°C and maintained under argon atmosphere, acryloylchloride (19.5 ml , 0.24 mol) was added dropwise. After complete addition (30 min), the reaction mixture was stirred at room temperature overnight, diluted with 300 ml of dichloromethane and washed with water (200 ml) and saturated NaCl solution (3 x 150 ml). The organic phase was dried over magnesium sulfate, filtered and evaporated to dryness. This gave 43 g of a brownish oil as a crude product. The crude product was filtered over a silica-gel column (Hexane/Ether : 19/1) to give 21.45 g of pure acrylic acid 6-chlorohexyl ester. The 6-chlorohexyl ester was dissolved in 500 ml of acetone containing sodium iodide (84.3 g , 0.56 mol) and heated at reflux for 22 h. The precipitated sodium chloride was filtered off and the solvent was removed to give a yellowish residue which was dissolved in 300 ml of ether and washed with water (100 ml), saturated NaCl solution (3 x 150 ml) and dried over magnesium sulfate. The magnesium sulfate was removed by filtration. Removal of the ether solvent gave acrylic acid 6-iodo-hexyl ester as yellowish oily compound. Yield 31.74 g.

20

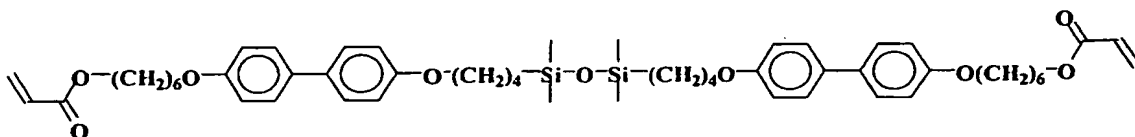
(b) Acrylic acid 6-(4'-hydroxybiphenyl-4-oxy)hexyl ester



25

A solution of 4, 4'-dihydroxybiphenyl (3.72 g , 20 mmol), acrylic acid 6-iodohexyl ester (4.21 g , 15 mmol) and potassium carbonate (2.07 g , 15 mmol) in DMF (50 ml) was stirred overnight at room temperature and under argon atmosphere. The reaction mixture was then poured into HCl 3*N* (100 ml) and extracted with ether (3 x 200 ml). The combined organic extracts were washed with saturated NaCl solution (200 ml), dried over magnesium sulfate, filtered and evaporated to dryness. The crystalline residue obtained was purified on silica-gel column (CH₂Cl₂, CH₂Cl₂/Et₂O : 19/1). This gave acrylic acid 6-(4'-hydroxybiphen-4-yloxy-)hexyl ester as white crystalline material. Yield 2.5 g.

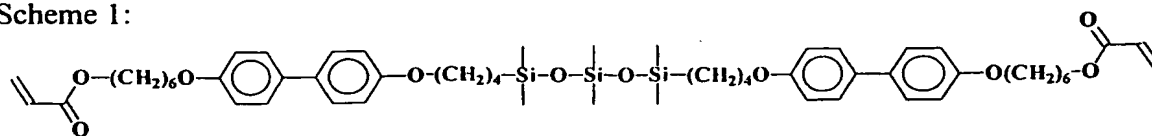
(c) 1,3-Bis[4-(4-oxo-4'-(6-acryloyloxyhexyloxy)biphenyl)but-1-yl]-1, 1, 3, 3-tetramethyldisiloxane



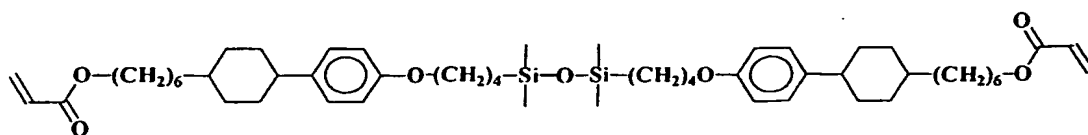
To a solution of acrylic acid 6-(4'-hydroxybiphen-4-yloxy-)hexyl ester (1.77 g , 5.2 mmol), triphenylphosphine (TPP) (1.36 g , 5.2 mmol) and 1, 3-bis(4-hydroxybutyl)-1, 1, 3, 3-tetramethyldisiloxane (0.70 g , 2.5 mmol) in dry THF (30 ml), cooled at 0°C and maintained under argon atmosphere, diethyl azodicarboxylate (0.91 g , 5.2 mmol) was added dropwise over 30 min. After complete addition the reaction mixture was stirred at room temperature for 6 hours, then evaporated to dryness to give a residue which was purified by chromatography over a silica-gel column (CH₂Cl₂) to afford 1,3-Bis[4-(4-oxo-4'-(6-acryloyloxyhexyloxy)biphenyl)but-1-yl]-1, 1, 3, 3-tetramethyldisiloxane as white crystalline material. Yield 2.1 g

This compound has the following thermotropic sequence: K 99°C I

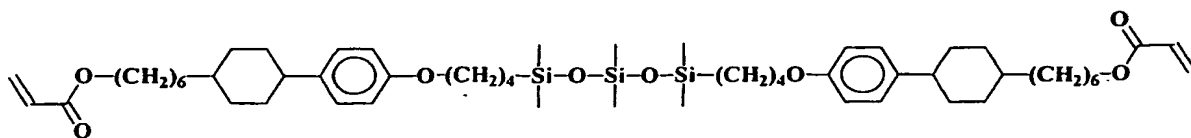
The following compounds may also be prepared in a similar way in accordance with Scheme 1:



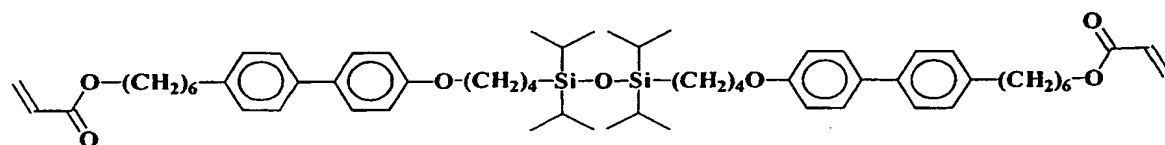
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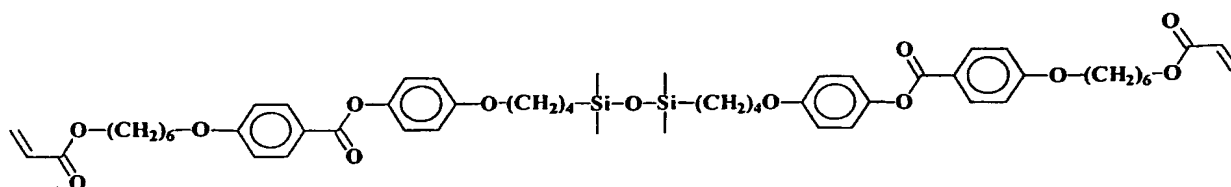
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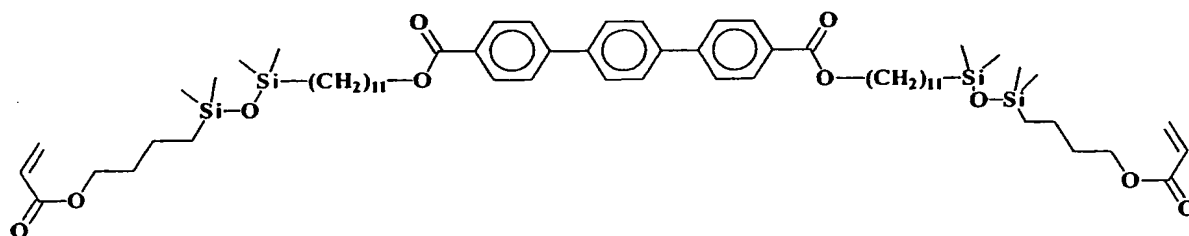
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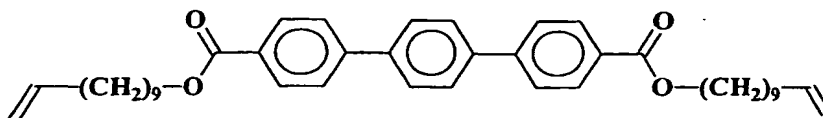
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Example 2:

25 [1,1' ; 4',1'']Terphenyl-4,4''-dicarboxylic acid bis-{11-[3-(4-acryloyloxybutyl)1,1,3,3-tetramethyldisiloxanyl]undecyl} ester



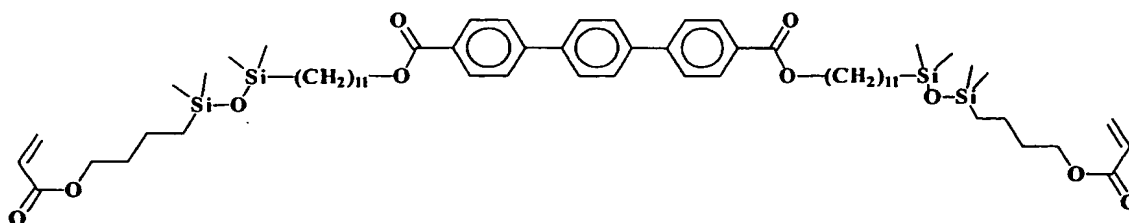
a) [1,1' ; 4',1'']Terphenyl-4,4''-dicarboxylic acid diundec-10-enyl ester



A suspension of [1,1' ; 4',1'']terphenyl-4,4''-dicarboxylic acid (6.37 g , 20 mmol) in 100
 5 ml of thionyl chloride was heated at reflux for 12 h then thionyl chloride was evaporated
 by azeotropic distillation. The crude yellow residue obtained therefrom was mixed with
 10-undecen-1-ol (7.49 g , 44 mmol) in dichloromethane (300 ml) and triethylamine
 (8.9 g , 88 mmol) was added dropwise to the resulting mixture. The reaction mixture was
 stirred at 6 h at room temperature, then poured into water (300 ml) and extracted with
 10 dichloromethane (2 x 400 ml). The combined organic extracts were washed with
 saturated NaCl solution (500 ml), dried over magnesium sulphate, filtered and evaporated
 to dryness to give a brownish residue which was recrystallised from acetone /
 dichlorometane: 9 / 1 to afford pure [1,1';4',1'']terphenyl-4,4''-dicarboxylic acid diundec-
 10-enyl ester as white crystalline material. Yield 8.97 g.

15

b) [1,1' ; 4',1'']Terphenyl-4,4''-dicarboxylic acid bis-{11-[3-(4-acryloyloxybutyl)-1,1,3,3-tetramethyldisiloxanyl]undecyl} ester



To a mixture of [1,1' ; 4',1'']terphenyl-4,4''-dicarboxylic acid diundec-10-enyl ester (5g,
 20 8.03 mmol) and 1,1,3,3-tetramethyldisiloxane (8.63g, 64.24 mmol.) in benzene (20 ml),
 stirred at room temperature and under argon atmosphere, 20 µl of platinum –
 divinyltetramethyldisiloxane catalyst [PT:DVTMS] (complex in xylene, 2.1-2.4%
 platinum) was added. The hydrosilylation reaction started immediately. After stirring for
 1 h at room temperature, further 5 µl of the platinum catalyst was added to the reaction
 25 mixture, which was then heated at 70°C overnight. A further 5 µl of the platinum catalyst

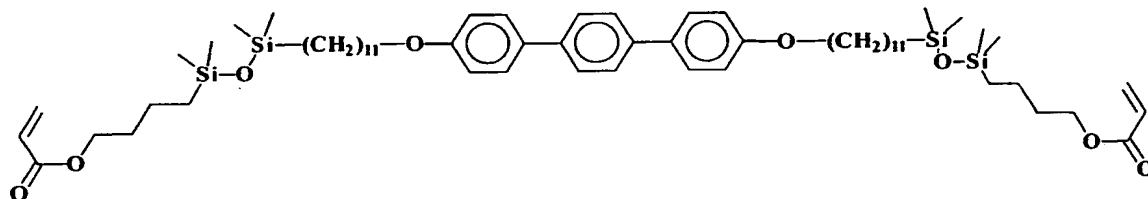
was added and the temperature was increased to 80°C. The reaction mixture was stirred for a further 1 h at this temperature. Removal of the benzene and excess disiloxane from the reaction mixture by distillation under reduced pressure gave a pasty yellow residue that was mixed with hot hexane (100 ml). The hexane mixture was filtered hot and evaporated to dryness. The pasty residue was dissolved in 40 ml of dry benzene and mixed with 3-buten-1-ol (2.31 g , 32 mmol)). The resulting solution was stirred for 15 min at room temperature. 10 µl of the platinum catalyst was then added to the solution and the reaction mixture was heated to 80°C for 2 h. A further 10 µl of the platinum catalyst was added to the reaction mixture which was stirred for a further 1 h at 80°C.

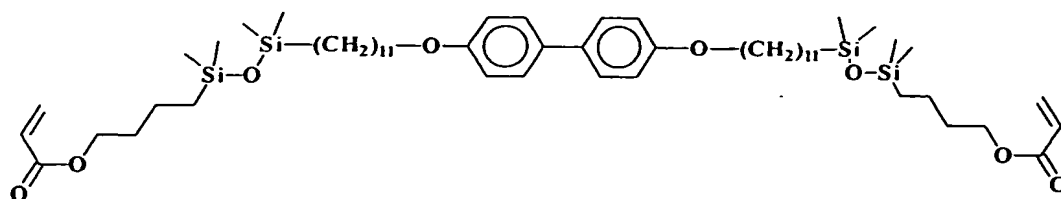
The resulting dark reaction mixture was evaporated to dryness to give a dark pasty residue which was dissolved in 120 ml of dry THF. To this mixture 10 ml of Et₃N was added followed by dropwise addition (at -10 to 0°C and under N₂) of a solution of acryloyl chloride (2.9 g , 32 mmol) in 30 ml of dry THF. After complete addition (20 min) the reaction mixture was stirred for further 6h at room temperature. THF and excess Et₃N were removed under reduced pressure to give a dark-yellow residue that was dissolved in minimum CH₂Cl₂ (80 ml) and poured into 300 ml of Et₂O. The precipitated triethylammonium salts were removed by filtration and the filtrate was evaporated to dryness to give a brown residue (14.2g) which was purified by chromatography on SiO₂ column using Et₂O / Hexane : 1 / 4 as eluent to give [1,1';4',1'']terphenyl-4,4"-dicarboxylic acid bis-{1,1-[3-(4-acryloyloxybutyl)-1,1,3,3-tetramethyldisiloxanyl]undecyl} ester as pasty white product. Yield 5.58 g.

This compound has the following thermotropic sequence: D 40°C I

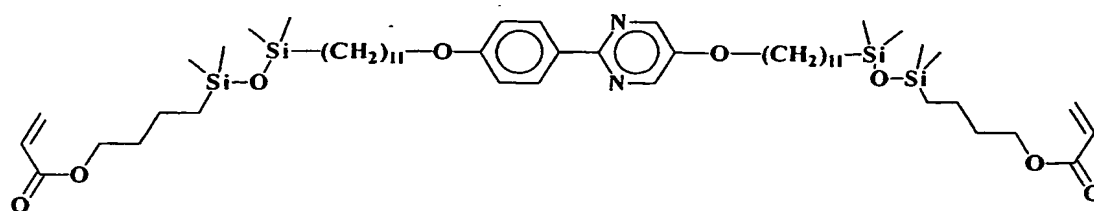
The following compounds may also be prepared in a similar way in accordance with

Scheme 2:

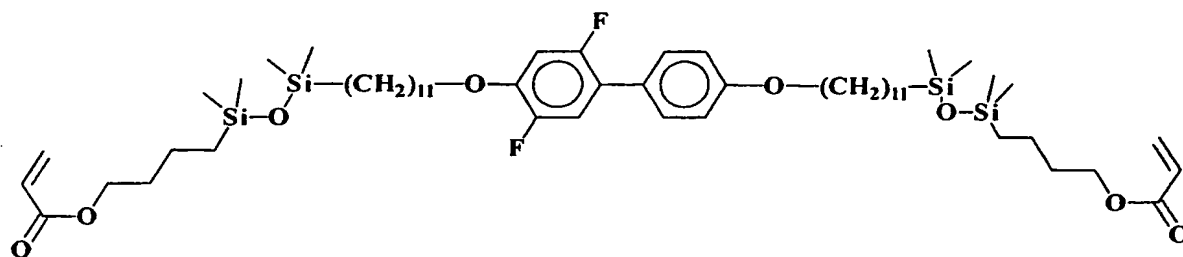




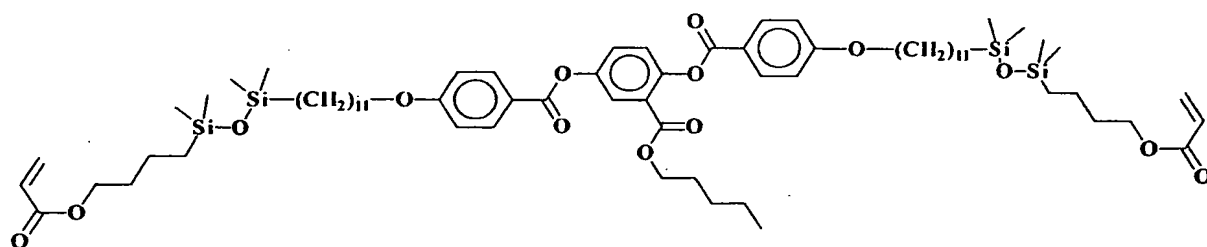
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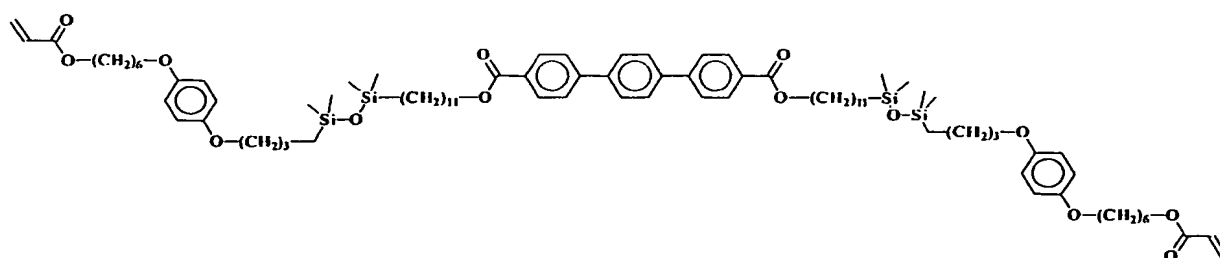
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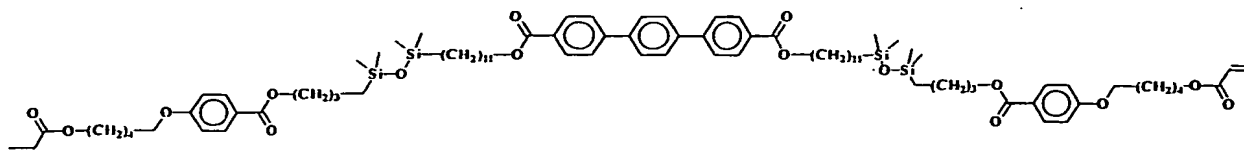
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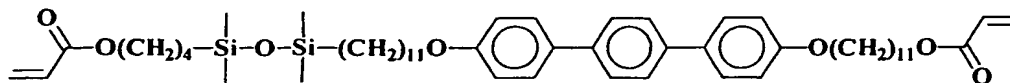
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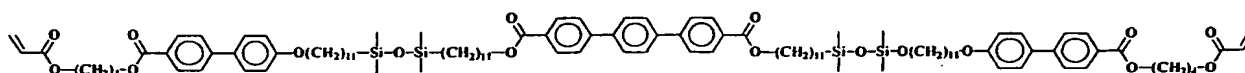
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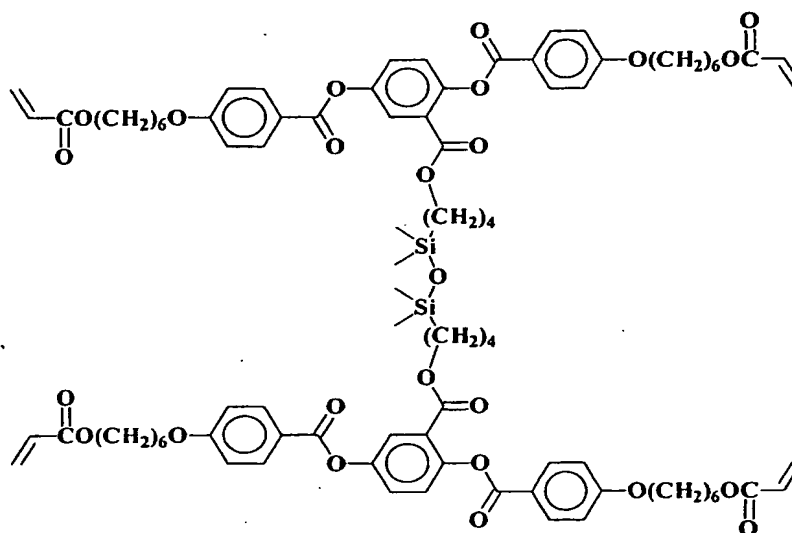
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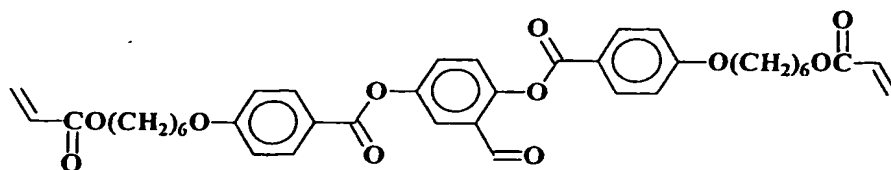
Example 3:

1,3-Bis-{4-[2,5-di-(4-(6-acryloyloxyhexyloxy)phenylcarbonyloxy)phenylcarbonyloxy]-but-1-yl}-1, 1, 3, 3-tetramethyldisiloxane



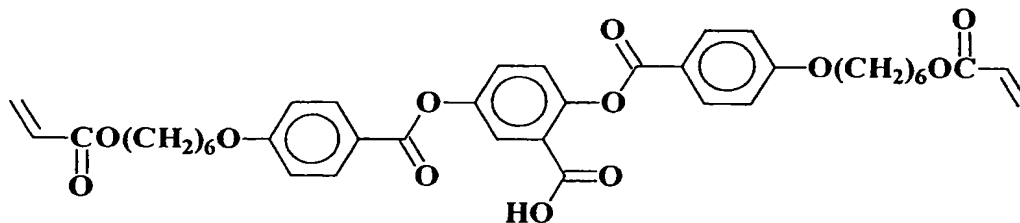
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a) 2,5-Di-[4-(6-acryloyloxyhexyloxy)phenylcarbonyloxy]benzaldehyde



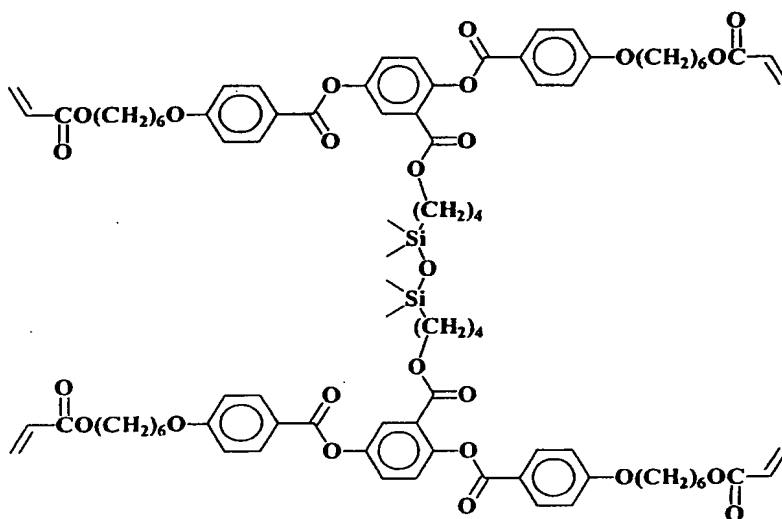
5 A solution of mesylchloride (4.23 g, 36.94 mmol) in 10 ml of dry THF was added dropwise under argon over a period of 15 minutes to a cooled (-25 °C) solution of 4-(6-acryloyloxyhexyloxy)benzoic acid and triethylamine (20 ml) in 80 ml of dry THF. The reaction mixture was then stirred for 60 min at -25 °C, treated with a solution of 2,5-dihydroxybenzaldehyde (2.3 g, 16.65 mmol) in 60 ml of dry THF containing 195 mg of DMAP and further stirred at -25°C for 2h. The reaction mixture was then allowed to warm to room temperature and stirring was continued overnight. The reaction mixture was then poured into 120 ml of saturated NaHCO₃ and extracted with 2 x 200 ml of ether. The combined organic extracts were washed with 3N HCl (200 ml) and semi-saturated NaCl solution (2 x 100 ml), dried over MgSO₄, filtered and dried to give a slightly yellow pasty material. This was purified by flash chromatography over a short silica column (CH₂Cl₂/Et₂O : 19.5 / 0.5) to give a white residue (9.25 g) which was dissolved in CH₂Cl₂ (25 ml) then recrystallised from ethanol (250 ml) to give pure 2,5-di-[4-(6-acryloyloxyhexyloxy)phenylcarbonyloxy]benzaldehyde as a white crystalline material. Yield 8.5 g.

b) 2,5-Di-[4-(6-acryloyloxyhexyloxy)phenylcarbonyloxy]benzoic acid



Jones oxidant ($\text{CrO}_3/\text{H}_2\text{SO}_4/\text{H}_2\text{O}$) (48 ml) was added to a ice-cooled solution of 2,5-di-[4-(6-acryloyloxyhexyloxy)phenylcarbonyloxy]benzaldehyde (8.24 g, 12 mmol) in acetone (300 ml) in a dropwise fashion over a period of 30 min. The reaction mixture was stirred overnight at room temperature. The resulting green-orange mixture was filtered off to leave a green precipitate that was washed with 600 ml of ether. The combined organic solutions were washed with water until the orange coloration disappeared (6 x 250 ml). The colourless organic solution obtained was washed with saturated NaCl solution (2 x 300 ml), dried over MgSO_4 and filtered. Removal of the solvent gave pure 2,5-di-[4-(6-acryloyloxyhexyloxy)phenylcarbonyloxy]benzoic acid as a white crystalline material. Yield 8.5 g.

c) 1,3-Bis-{4-[2,5-di-(4-(6-acryloyloxyhexyloxy)phenylcarbonyloxy)-phenylcarbonyloxy]but-1-yl}-1, 1, 3, 3-tetramethyldisiloxane

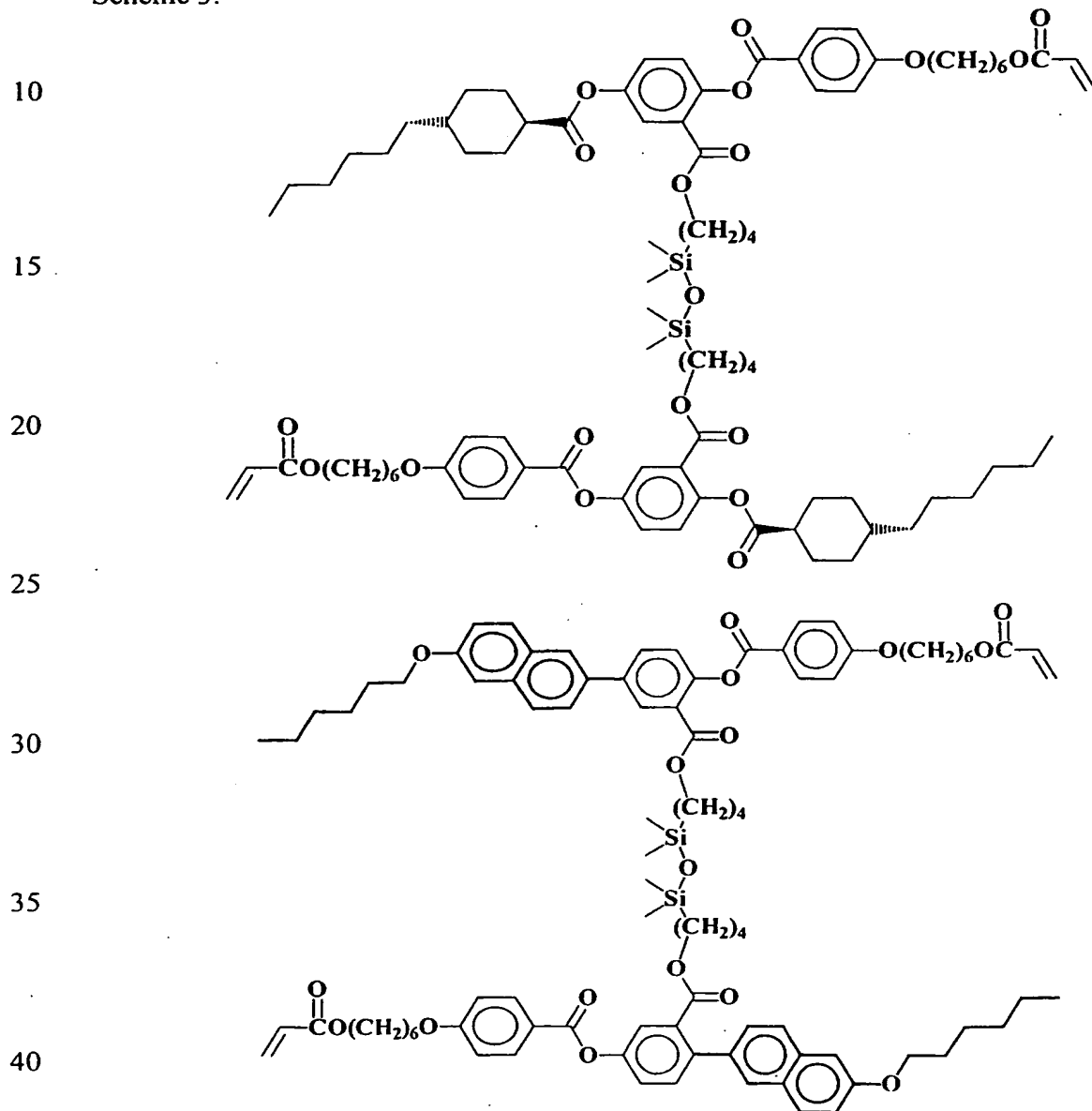


To a stirred solution of 2,5-di-[4-(6-acryloyloxyhexyloxy)phenylcarbonyloxy]-benzoic acid (0.77 g, 1.10 mmol), 4-dimethylaminopyridine (13 mg, 0.11 mmol) and 1,3-bis(4-hydroxybutyl)-1, 1, 3, 3-tetramethyldisiloxane (0.14 g, 0.5 mmol) in dichloromethane (25 ml), at 0°C , a solution of $\text{N,N}'$ -dicyclohexylcarbodiimide (0.21 g, 1.10 mmol) in dichloromethane (5 ml) was added dropwise over a period of 15 min. The reaction mixture was warmed to room temperature and stirred overnight. Removal of the

solvent gave a residue which was then purified by chromatography over a short silica-gel column (CH_2Cl_2 , $\text{CH}_2\text{Cl}_2/\text{Et}_2\text{O}$: 19 / 1) to give 1,3-bis-{4-[2,5-di-(4-(6-acryloyloxyhexyloxy)phenylcarbonyloxy)phenylcarbonyloxy]-but-1-yl}-1,3-tetramethyldisiloxane as a white crystalline material. Yield: 0.59g.

5 This compound has the following thermotropic sequence: N 43.8°C I

The following compounds may also be prepared in a similar way in accordance with Scheme 3:



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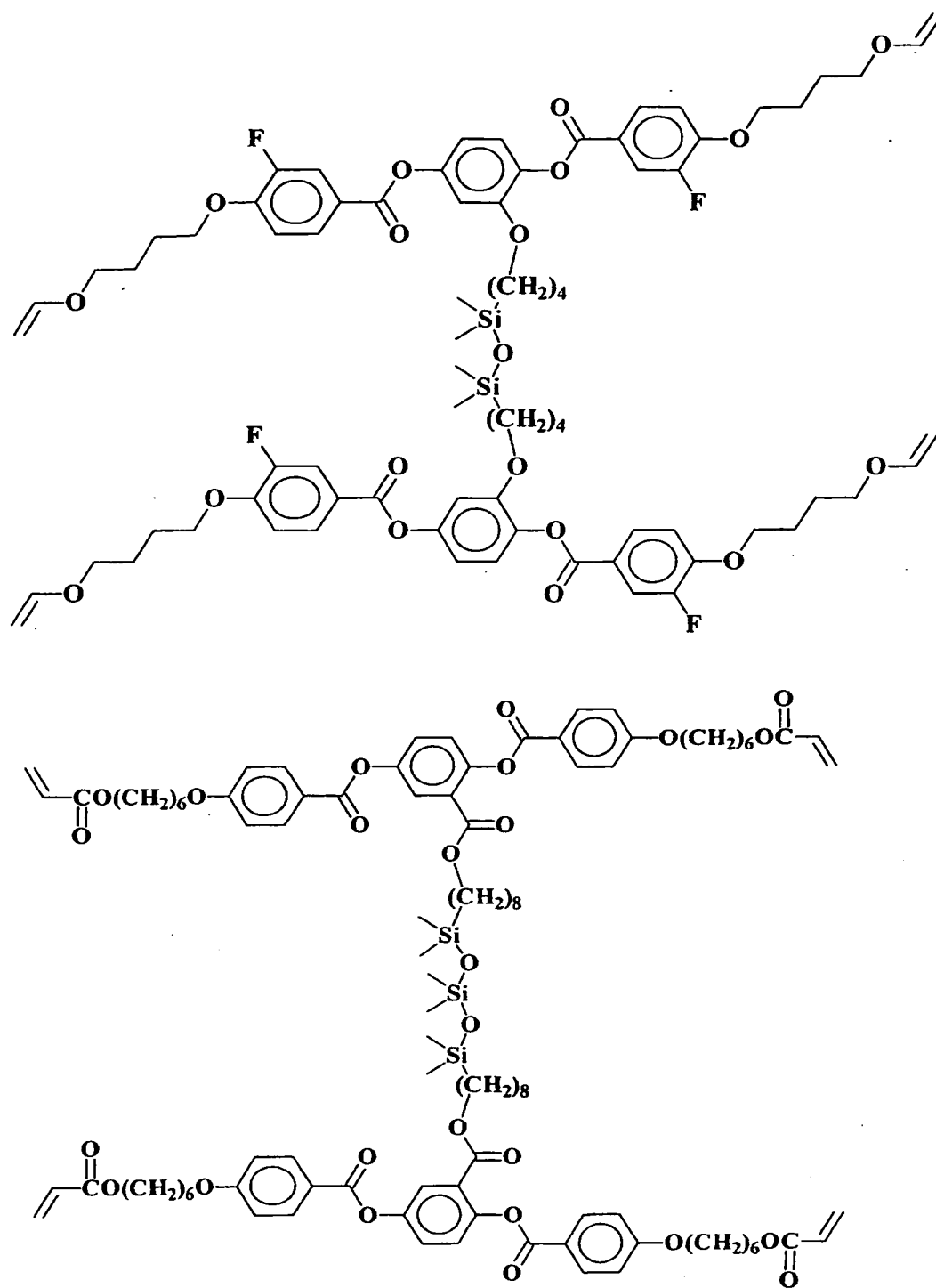
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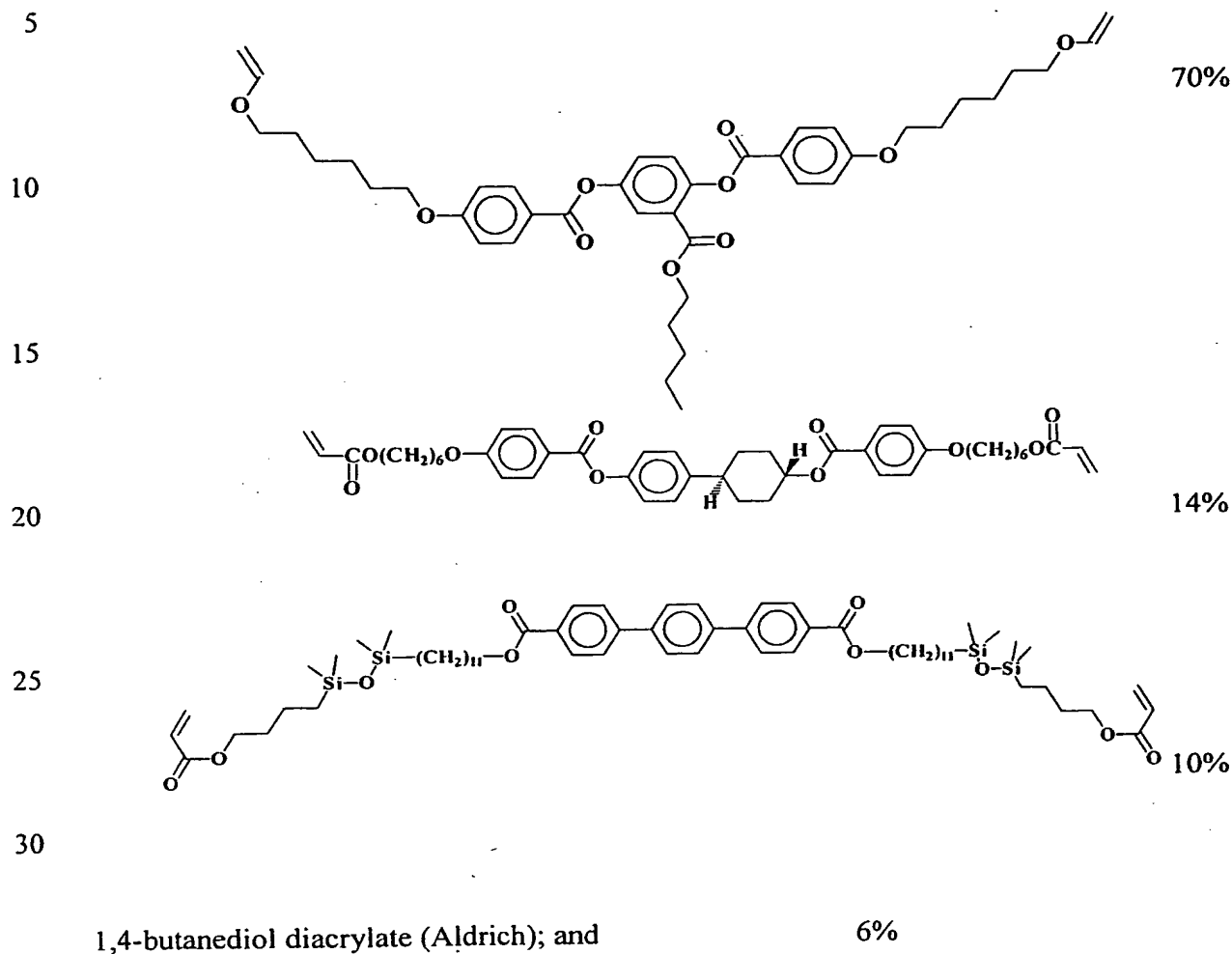
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Example 4Preparation of Nematic LCP Films

A mixture of the following components in Anisole was prepared:



500ppm of 2,6-di-(t-butyl)-4-hydroxytoluene (BHT) inhibitor were added to this mixture in order to prevent premature polymerization. Polymerization was started using 500 ppm initiator such as Irgacure 369 (commercially available from Ciba Geigy, Basel, Switzerland). The mixture was stirred at room temperature and then spin-coated on a glass plate having an orientation layer to form an LCP film of ca.800nm in thickness. This film was dried at 60°C for 1 or 2 minutes and photopolymerized by irradiation with

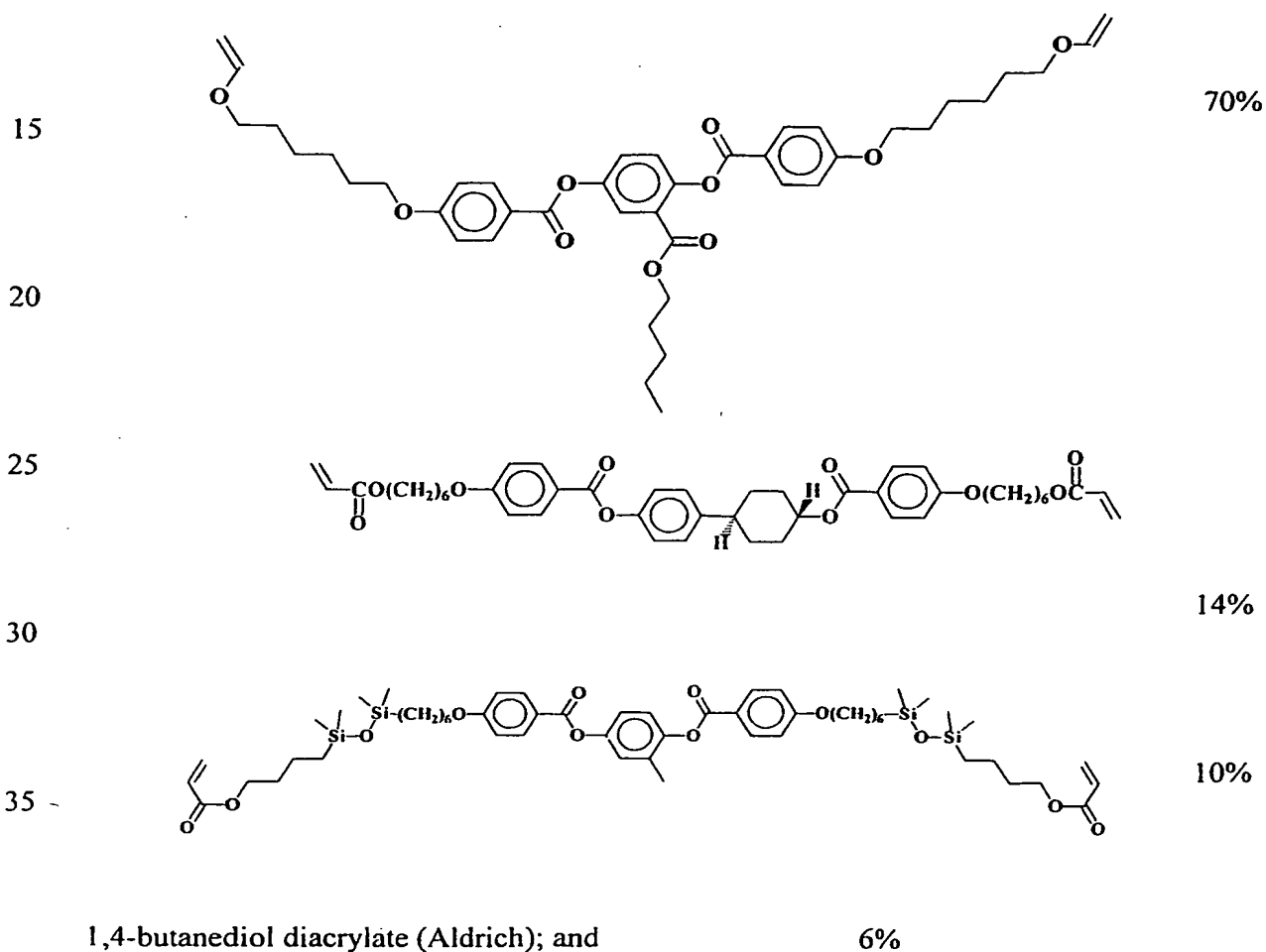
UV light for approximately 5 minutes at room temperature in an N₂-atmosphere using a Mercury lamp.

The well oriented film shows the nematic mesophase at room temperature with a clearing point of 85°C. In addition this film exhibits a tilt angle of about 30° relative to the plane of the substrate, as shown by ellipsometric measurements.

Example 5

Preparation of Nematic LCP Films

A mixture of the following components in anisole was prepared according to the procedure of Example 4:

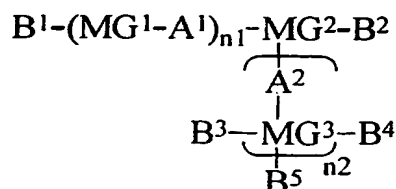


The well oriented film shows the nematic mesophase at room temperature with a clearing point of 86.5°C. In addition this film exhibits a tilt angle of about 30° relative to the plane of the substrate, as shown by ellipsometric measurements.

5

CLAIMS

1. Use of a component of a LC mixture for the manufacture of a LCP network comprising the mixture, characterised in that the component induces a stable tilt angle in the LCP network.
2. Use according to claim 1, in which the component comprises a substrate associating portion and a mesogenic extending portion characterised in that the lateral dimensions of the substrate associating group are at least half as large again as the lateral dimensions of the mesogenic extending group.
3. Use according to Claim 1 or Claim 2, in which the substrate associating group comprises an organosiloxane group.
4. Use according to any one of claims 1 to 3, in which the mesogenic extending group comprises an aromatic, a non-aromatic carbocyclic or heterocyclic ring system
5. Use according to any one of claims 1 to 4, in which the LCP network is a nematic, smectic A or discotic material.
6. A compound of formula (I)



(I)

wherein

B¹ to B⁴ each individually represent a spacer group selected from the group comprising optionally substituted saturated or unsaturated, straight chain or branched chain C₂₋₈₀ alkyl group, one or more of the alkyl CH₂ groups being optionally replaced by one or more heteroatoms;

B⁵ represents a hydrogen atom or spacer group as defined above,

A¹ and A² each individually represent a single bond or a spacer group as defined above.

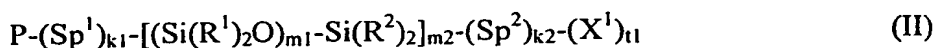
MG¹ to MG³ each individually represent a mesogenic group comprising at least one optionally substituted aromatic or non aromatic carbocyclic or heterocyclic ring systems; and

n₁ and n₂ are each independently 0 or a positive integer

with the proviso that, firstly, when n₂ is 0 both B¹ and B² include a polymerisable group and at least one of A¹, B¹ and B² includes a group selected from an organosiloxane, an organogermanium, an organotin and an organo-perfluoro residue and

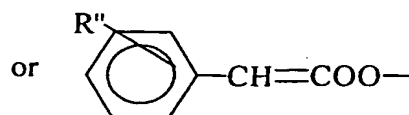
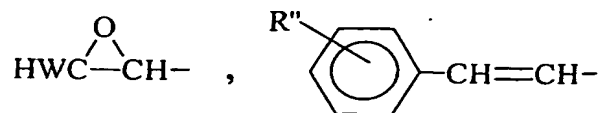
secondly, when n₂ > 0 at least one of B¹, B², B³ and B⁴ includes a polymerisable group and at least one of A¹, A², B¹, B², B³ and B⁴ includes an organosiloxane group.

7. A compound according to claim 6, in which at least one of A¹, B¹ and B² includes an organosiloxane group.
8. A compound according to Claim 6 or Claim 7, in which at least one of the groups B₁ to B₄ includes a bulky group selected from a fluorinated alkyl residue and an organic residue including an atom selected from germanium, tin and silicon.
9. A compound according to anyone of claims 6 to 8, in which at least one of the groups B₁ to B₄ include at least one siloxane group.
10. A compound according to Claim 8 or Claim 9, in which the groups B₁ and B₂ include at least one siloxane group.
11. A compound according to any one of claims 6 to 10, in which B¹ to B⁴ comprise a group of formula (II)



wherein

P is a polymerisable group selected from groups comprising CH₂=CW-, CH₂=CW-O-, CH₂=CW-COO-, CH₂=C(Ph)-COO-, CH₂=CH-COO-Ph-, CH₂=CW-CO-NH-, CH₂=C(Ph)-CONH-, CH₂=C(COOR')-CH₂-COO-, CH₂=CH-O-, CH₂=CH-OOC-, Ph-CH=CH-, CH₃-C=N-(CH₂)_{m3}-, HO-, HS-, HO-(CH₂)_{m3}-, HS-(CH₂)_{m3}-, HO(CH₂)_{m3}COO-, HS(CH₂)_{m3}COO-, HWN-, HOC(O)-, CH₂=CH-Ph-(O)_{m4}



wherein

W represents H, Cl or a C₁₋₅ alkyl group;

m₃ is an integer having a value of from 1 to 9;

m₄ is an integer having a value of 0 or 1,

R' represents a C₁₋₅ alkyl group;

R'' represents a C₁₋₅ alkyl group, methoxy, cyano, F, Cl, Br or I;

Sp¹ and Sp² each independently represent a C₁₋₂₀ alkyl group;

k₁ and k₂ are integers, each independently having a value of from 0 to 4;

R¹ and R² each independently represent straight chain or branched C₁₋₂₀ alkyl groups;

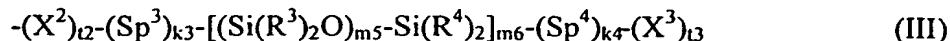
m₁ is an integer having a value of from 0 to 10;

m₂ is an integer having a value of from 1 to 10;

X¹ represents -O-, -S-, -NH-, N(CH₃)-, -CH(OH)-, -CO-, -CH₂(CO)-, -SO-, -CH₂(SO)-, -SO₂-, -CH₂(SO₂)-, -COO-, -OCO-, -OCO-O-, -SCO-, -CO-S-, -SOO-, -OSO-, -SOS-, -CH₂-CH₂-, -OCH₂-, -CH₂O-, -CH=CH-, -C≡C- or a single bond; and

t₁ is an integer having a value of 0 or 1.

12. A compound according to any one of claims 6 to 11, in which A¹ and/or A² comprise a group of formula (III)



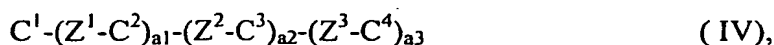
wherein

Sp³ and Sp⁴ each independently represent the designations assigned to Sp¹ above,

- X^2 and X^3 each independently represent the designations assigned to X^1 above,
 R^3 and R^4 each independently represent the designations assigned to R^1 above,
 5 k_3 and k_4 are integers, each integer independently having a value of from 0 to 4,
 t_4 and t_3 are integers, each independently having a value of 0 or 1;
 m_5 is an integer having a value of from 0 to 10; and
 m_6 is an integer having a value of from 1 to 10.

10

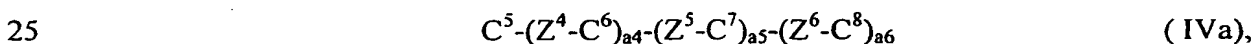
13. A compound according to any one of claims 6 to 12, in which MG^1 comprises a group of formula (IV)



wherein

- 15 C^1 to C^4 each independently represent a non aromatic, aromatic, carbocyclic or heterocyclic group;
 Z^1 to Z^3 each represent, independently, $-COO-$, $-OCO-$, $-CH_2-CH_2-$, $-OCH_2-$, $-CH_2O-$, $-CH=CH-$, $-C\equiv C-$, $-CH=CH-COO-$, $-OCO-CH=CH-$ or a single bond; and
 20 a_1 , a_2 and a_3 are integers, each integer independently having a value of from 0 to 3, with the proviso that $a_1 + a_2 + a_3 \leq 3$.

14. A compound according to any one of claims 6 to 13, in which MG^2 and MG^3 comprise a group of formula (IVa)



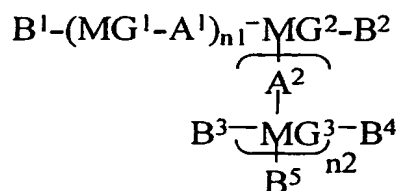
in which:

- C^5 , C^6 , C^7 and C^8 each independently represent a non aromatic, aromatic, carbocyclic or heterocyclic group;
 Z^4 to Z^6 each independently represent $-COO-$, $-OCO-$, $-CH_2-CH_2-$, $-OCH_2-$, $-CH_2O-$, $-CH=CH-$, $-C\equiv C-$, $-CH=CH-COO-$, $-OCO-CH=CH-$ or a single bond; and
 30

a4, a5 and a6 are integers, each integer independently having a value of from 0 to 3,

with the proviso firstly that $a4 + a5 + a6 \leq 3$ and secondly that at least one of C⁵, C⁶, C⁷ and C⁸ independently comprises a trivalent group.

5 15. A compound of formula (I) according to any one of claims 6 to 14,



10

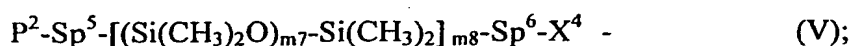
in which

n1 and n2 are integers, each independently having a value of 0 or 1;

B⁵ is hydrogen;

B¹ to B⁴ each independently represent a group of formula (V)

15



A1 and A2 each independently represent a group of formula (VI)

20



wherein

X⁴ to X⁶ each independently represent -O-, -CO-, -COO-, -OCO-, -C≡C-, or a single bond, especially -O-, -COO-, -OCO- or single bond;

25

Sp⁵ to Sp⁸ each independently represent a C₁₋₂₀ straight-chains alkyl group, especially ethylene, propylene, butylene, pentylene, hexylene, heptylene, octylene, nonylene, decylene, undecylene, or dodecylene;

30

P² represents CH₂=CW⁵- or CH₂=CW⁵-(CO)_{v2}O- wherein W⁵ represents H, CH₃, or Cl; and

v_2 is an integer, having a value of 0 or 1;

m_7 and m_9 are integers, each integer independently having a value of 1 or 2;

m_8 and m_{10} are integers, each integer independently having a value of 0 or 1, with the proviso that $m_8 + m_{10} > 0$;

MG^1 represents a group of formula IV

wherein

C^1 , C^2 , C^3 and C^4 are the same and are selected from phenylene or biphenylene; and

Z^1 to Z^3 each independently represent a single bond, -COO- or -OCO-; and

MG^2 and MG^3 independently represent a group of formula IVa

wherein

C^5 , C^6 , and C^8 are the same and are selected from phenylene or biphenylene;

C^7 represents phenylene or biphenylene; and

Z^4 to Z^6 each independently represent a single bond, -COO- or -OCO-.

16. A compound of formula (I) according to any one of claims 6 to 15

in which

n_1 and n_2 are both 0,

B_1 and B_2 each independently represent a group of formula (VII)

$P^3-Sp^5-[Si(CH_3)_2O]_{m_7}-Si(CH_3)_2]_{m_8}-Sp^6-X^4-$ (VII);

wherein

X^4 represents -O-, -CO-, -COO-, -OCO-, -C≡C- or a single bond, especially -O-, -COO-, -OCO- or a single bond;

Sp^5 to Sp^6 each independently represent a C_{1-20} straight-chain alkyl group, especially ethylene, propylene, butylene, pentylene, hexylene, heptylene, octylene, nonylene, decylene, undecylene, or dodecylene;

P^3 represents a hydrogen or CH_2-CW^5- or $CH_2-CW^5-(CO)_{v_2}O-$

wherein

W^5 represents H, CH₃, F, Cl, Br or I; and

v_2 is 0 or 1;

m_7 is 1 or 2;

5 m_8 is 1;

MG^2 represents a group of formula IVa

wherein

C^5, C^6 , and C^8 are the same and are selected from phenylene or biphenylene;

C^7 represents phenylene or biphenylene; and

10 Z^4 to Z^6 each independently represent a single bond, -COO-, or -OCO-.

17. A compound of formula (I) being 1,3-bis-{4-[4-oxo-4'-(6-acryloyloxy-hexyloxy)biphenyl]but-1-yl}-1, 1, 3, 3-tetramethyldisiloxane.

18. A compound of formula (I) being [1,1'; 4',1'']terphenyl-4,4''-dicarboxylic acid bis-{1,1-[3-(4-acryloyloxybutyl)1,1,3,3-tetramethyldisiloxanyl]undecyl} ester.

15 19. A compound of formula (I) being 1,3-bis-{4-[2,5-di-(4-(6-acryloyloxy-hexyloxy)phenylcarbonyloxy)phenylcarbonyloxy]but-1-yl}-1, 1, 3, 3-tetramethyldisiloxane.

20. A method for the preparation of a compound according to any one of claims 6 to 19 comprising reacting a mesogenic compound with a directive siloxane.

20 21. A liquid crystalline mixture comprising at least two components, wherein at least one component is a compound of formula (I) as defined in any one of claims 6 to 19.

22. A liquid crystalline material comprising a compound of formula (I) as defined in any one of claims 6 to 19 or a liquid crystalline mixture according to Claim 21.

25 23. A liquid crystal polymer film comprising a compound of formula (I) as defined in any one of claims 6 to 19 or a mixture as defined in Claim 21.

24. A material or film according to Claim 22 or 23, characterised in that the film or material is nematic, smectic A or discotic.

25. A liquid crystal device comprising a compound of formula (I) as defined in any

one of claims 6 to 19, a liquid crystalline material according to Claim 22, a liquid crystalline mixture according to Claim 21 or a liquid crystal polymer film according to Claim 23.

- 5 26. Use of a compound according to any one of Claims 6 to 19 in a liquid crystal mixture for inducing a stable tilt angle in the liquid crystal mixture.
27. A discotic LC material comprising one or more compounds of formula (I) according to any one of claims 6 to 19.

INTERNATIONAL SEARCH REPORT

International Application No

PCT/IB 99/01287

A. CLASSIFICATION OF SUBJECT MATTER

IPC 6 C09K19/56 C09K19/40

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC 6 C09K C08G G02F

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	US RE35462 E (W.HAAS) 25 February 1997 see column 1, line 37 - column 5, line 28 see column 10, line 8 - line 25 ---	1,3,6, 11-15, 20,21,26
X	GB 2 317 186 A (K.K.TOSHIBA) 18 March 1998 see page 2, line 7 - page 4, line 4 see page 6, line 13 - line 27 --- -/--	1,3,6, 11-15, 20,21,26

☒ Further documents are listed in the continuation of box C.☒ Patent family members are listed in annex.

* Special categories of cited documents:

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Date of the actual completion of the international search

31 August 1999

Date of mailing of the international search report

08/09/1999

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INTERNATIONAL SEARCH REPORT

Inter. natl Application No

PCT/IB 99/01287

C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT		
Category	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
Y	ROBINSON W K ET AL: "PRELIMINARY COMMUNICATION BI-MESOGENIC ORGANOSILOXANE LIQUID CRISTAL MATERIALS EXHIBITING ANTIFERROELECTRIC PHASES" LIQUID CRYSTALS, vol. 23, no. 2, August 1997, pages 309-312, XP000701497	1,3,17
X	see the whole document	6,11-15
X	DIAZ F ET AL: "SYNTHESIS AND CHARACTERIZATION OF TWIN MESOGENS CONTAINING SILOXANEUNITS AS CENTRAL SPACERS" LIQUID CRYSTALS, vol. 16, no. 1, 1 January 1994, pages 105-113, XP000418464 see the whole document	6,11-15, 20
Y	COLES H J ET AL: "NEW LOW MOLAR MASS ORGANOSILOXANES WITH UNUSUAL FERROELECTRIC PROPERTIES" LIQUID CRYSTALS, vol. 15, no. 5, 1 November 1993, pages 739-744, XP000414623 see the whole document	1,3
Y	CORSELLIS E ET AL: "STRUCTURAL CHARACTERIZATION OF MONO-AND DI-MESOGENIC ORGANOSILOXANES: THE IMPACT OF SILOXANE CONTENT ON BIPHENYL BENZOATE SYSTEMS" LIQUID CRYSTALS, vol. 23, no. 2, August 1997, pages 235-239, XP000701491	1,3,17
X	see the whole document	6,11-15
Y	OZCAYIR Y ET AL: "MESOMORPHIC PROPERTIES OF MAIN CHAIN THERMOTROPIC PLCS AND MODEL COMPOUNDS CONTAINING A TETRAMETHYLDISILOXANE MOIETY IN THE SPACER" MOLECULAR CRYSTALS AND LIQUID CRYSTALS (INC. NONLINEAR OPTICS), vol. 185, 1 August 1990, pages 75-87, XP000162926	1,3,17
X	see the whole document	6,11-15

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INTERNATIONAL SEARCH REPORT

information on patent family members

International Application No

PCT/IB 99/01287

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
US RE35462 E	25-02-1997	DE 3920509 A	10-01-1991
		CA 2019380 A,C	22-12-1990
		CA 2019558 A,C	22-12-1990
		DE 59010209 D	25-04-1996
		DE 59010756 D	02-10-1997
		EP 0405326 A	02-01-1991
		EP 0404140 A	27-12-1990
		JP 3048688 A	01-03-1991
		JP 3034987 A	14-02-1991
		US 5158702 A	27-10-1992
		US 5106530 A	21-04-1992
		US 5277838 A	11-01-1994
GB 2317186 A	18-03-1998	JP 10114893 A	06-05-1998